

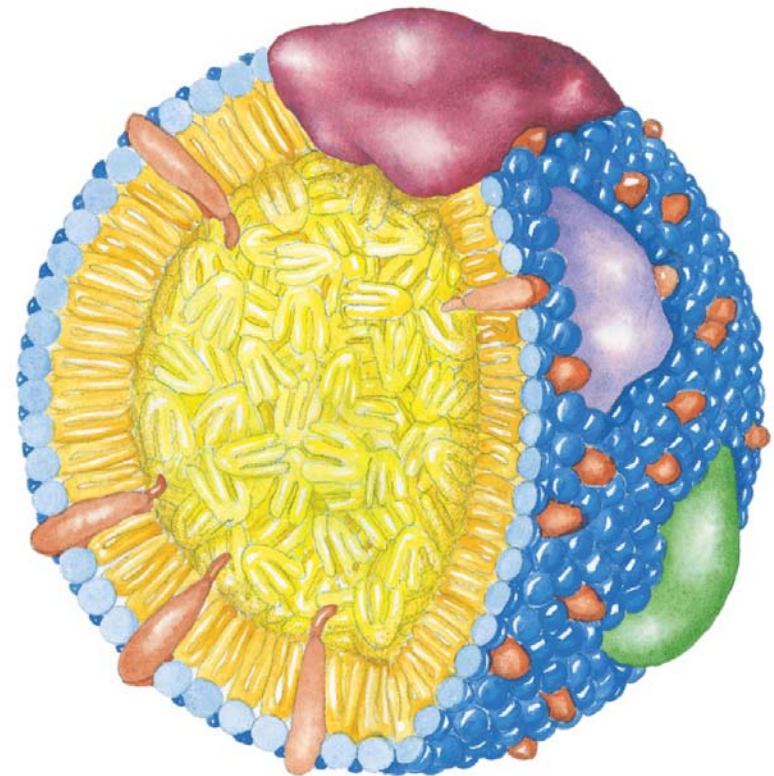
Chpt. 17

Fatty Acid Catabolism

17.1 Digestion, Mobilization, and Transport of Fats

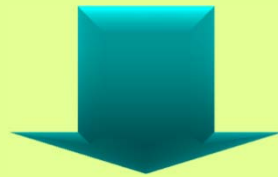
17.2 Oxidation of Fatty Acids

17.3 Ketone Bodies



Lipids

- Fats or Oils:** stored forms of energy
- Phospholipids, Sterols:** major structural elements of biological membrane
- Other lipids:** roles as enzyme cofactor, electron carrier, hydrophobic anchor, hormones, intracellular messengers



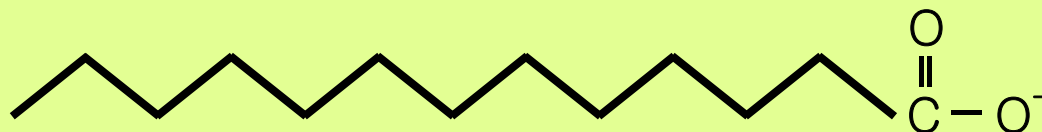
Storage lipids

Structural lipids in membranes

Lipids as signals, cofactors, and pigments

Fatty acids

Fatty acids: carboxylic acids with hydrocarbon chains (C₄ to C₃₆)



Saturated fatty acids
Unsaturated fatty acids

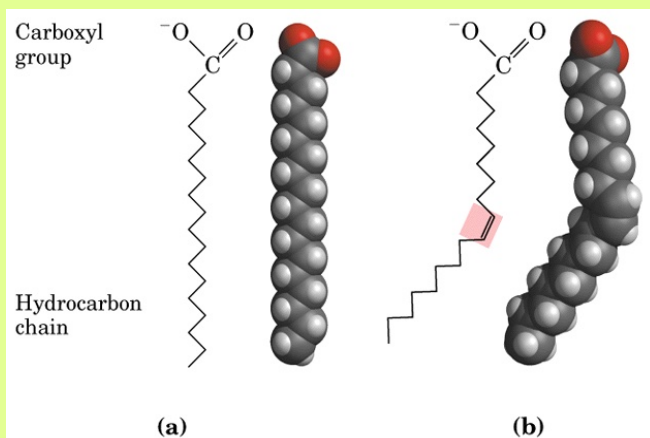


TABLE 10-1 Some Naturally Occurring Fatty Acids: Structure, Properties, and Nomenclature

Carbon skeleton	Structure*	Systematic name†	Common name (derivation)	Melting point (°C)	Solubility at 30 °C (mg/g solvent)	
					Water	Benzene
12:0	CH ₃ (CH ₂) ₁₀ COOH	<i>n</i> -Dodecanoic acid	Lauric acid (Latin <i>laurus</i> , "laurel plant")	44.2	0.063	2,600
14:0	CH ₃ (CH ₂) ₁₂ COOH	<i>n</i> -Tetradecanoic acid	Myristic acid (Latin <i>Myristica</i> , nutmeg genus)	53.9	0.024	874
16:0	CH ₃ (CH ₂) ₁₄ COOH	<i>n</i> -Hexadecanoic acid	Palmitic acid (Latin <i>palma</i> , "palm tree")	63.1	0.0083	348
18:0	CH ₃ (CH ₂) ₁₆ COOH	<i>n</i> -Octadecanoic acid	Stearic acid (Greek <i>stear</i> , "hard fat")	69.6	0.0034	124
20:0	CH ₃ (CH ₂) ₁₈ COOH	<i>n</i> -Eicosanoic acid	Arachidic acid (Latin <i>Arachis</i> , legume genus)	76.5		
24:0	CH ₃ (CH ₂) ₂₂ COOH	<i>n</i> -Tetracosanoic acid	Lignoceric acid (Latin <i>lignum</i> , "wood" + <i>cera</i> , "wax")	86.0		
16:1(Δ ⁹)	CH ₃ (CH ₂) ₅ CH=CH(CH ₂) ₇ COOH	<i>cis</i> -9-Hexadecenoic acid	Palmitoleic acid	1–0.5		
18:1(Δ ⁹)	CH ₃ (CH ₂) ₇ CH=CH(CH ₂) ₇ COOH	<i>cis</i> -9-Octadecenoic acid	Oleic acid (Latin <i>oleum</i> , "oil")	13.4		
18:2(Δ ^{9,12})	CH ₃ (CH ₂) ₄ CH=CHCH ₂ CH=CH(CH ₂) ₇ COOH	<i>cis</i> -, <i>cis</i> -9,12-Octadecadienoic acid	Linoleic acid (Greek <i>linon</i> , "flax")	1–5		
18:3(Δ ^{9,12,15})	CH ₃ CH ₂ CH=CHCH ₂ CH=CHCH ₂ CH=CH(CH ₂) ₇ COOH	<i>cis</i> -, <i>cis</i> -, <i>cis</i> -9,12,15-Octadecatrienoic acid	α-Linolenic acid	–11		
20:4(Δ ^{5,8,11,14})	CH ₃ (CH ₂) ₄ CH=CHCH ₂ CH=CHCH ₂ CH=CHCH ₂ CH=CH(CH ₂) ₃ COOH	<i>cis</i> -, <i>cis</i> -, <i>cis</i> -, <i>cis</i> -5,8,11,14-Icosatetraenoic acid	Arachidonic acid	–49.5		

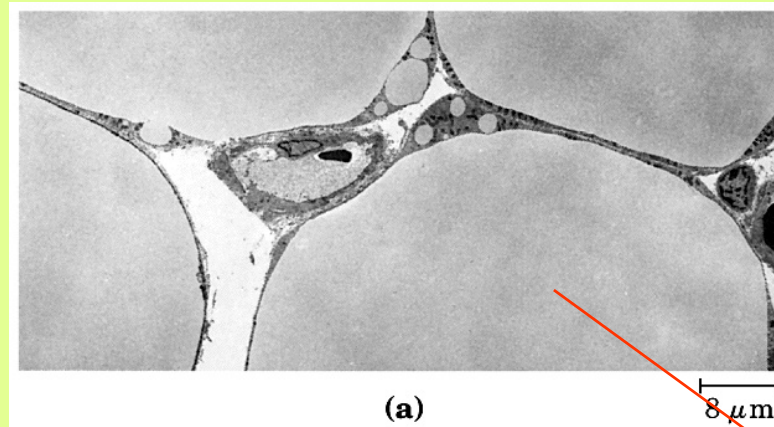
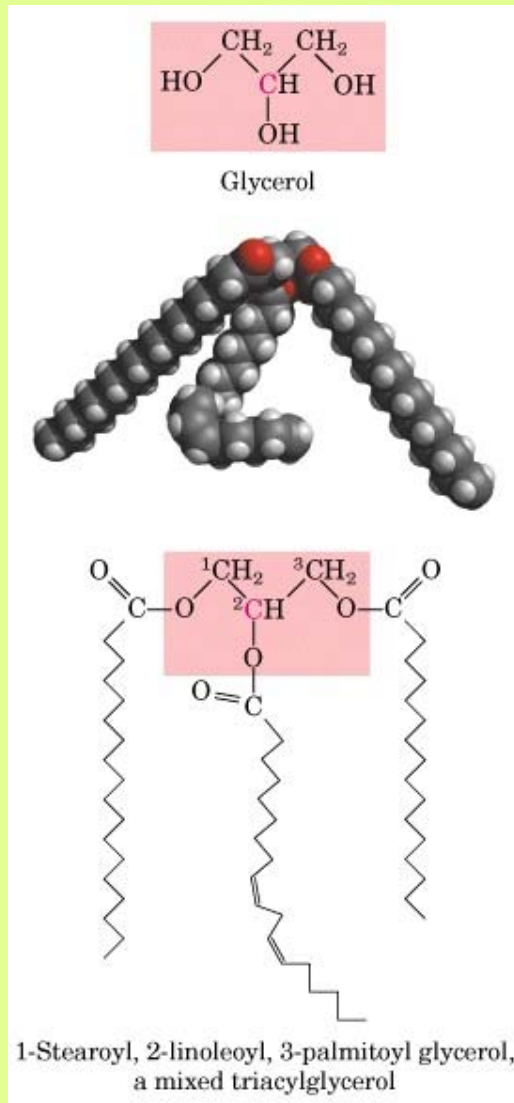
*All acids are shown in their nonionized form. At pH 7, all free fatty acids have an ionized carboxylate. Note that numbering of carbon atoms begins at the carboxyl carbon.

†The prefix *n*- indicates the "normal" unbranched structure. For instance, "dodecanoic" simply indicates 12 carbon atoms, which could be arranged in a variety of branched forms; "*n*-dodecanoic" specifies the linear, unbranched form. For unsaturated fatty acids, the configuration of each double bond is indicated; in biological fatty acids the configuration is almost always *cis*.

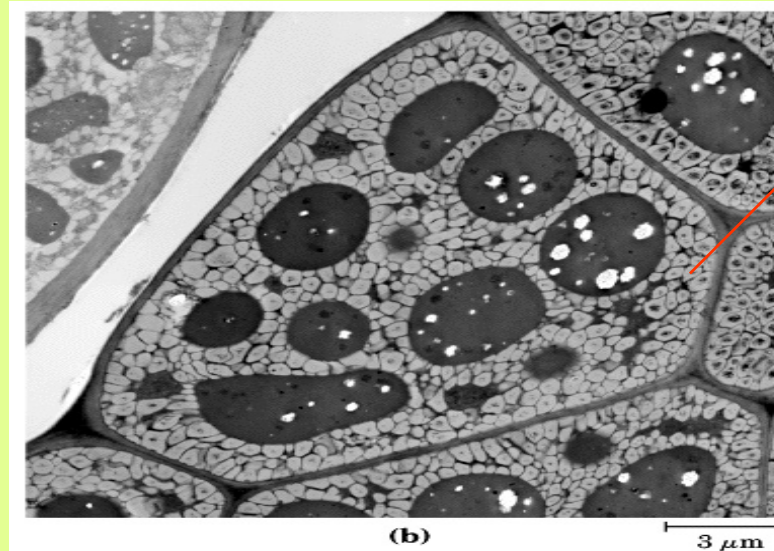
Triacylglycerol

Storage lipids (derivatives of fatty acids)

provide stored energy and insulation

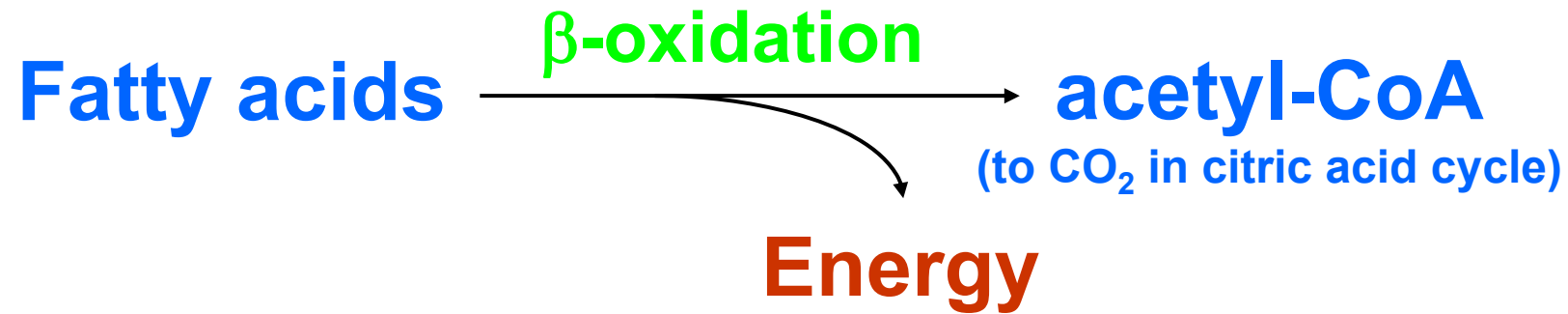


adipocytes



fat

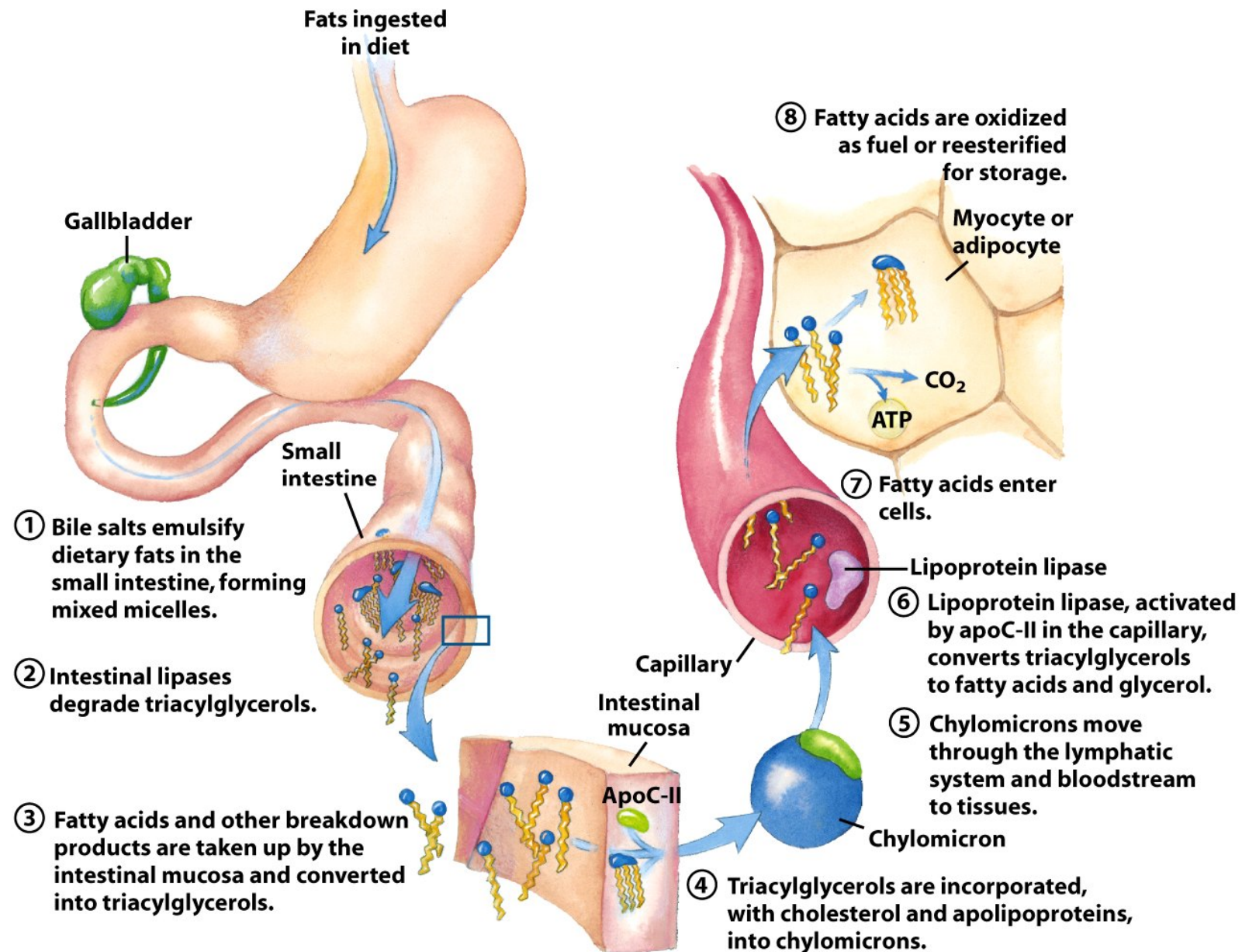
seed



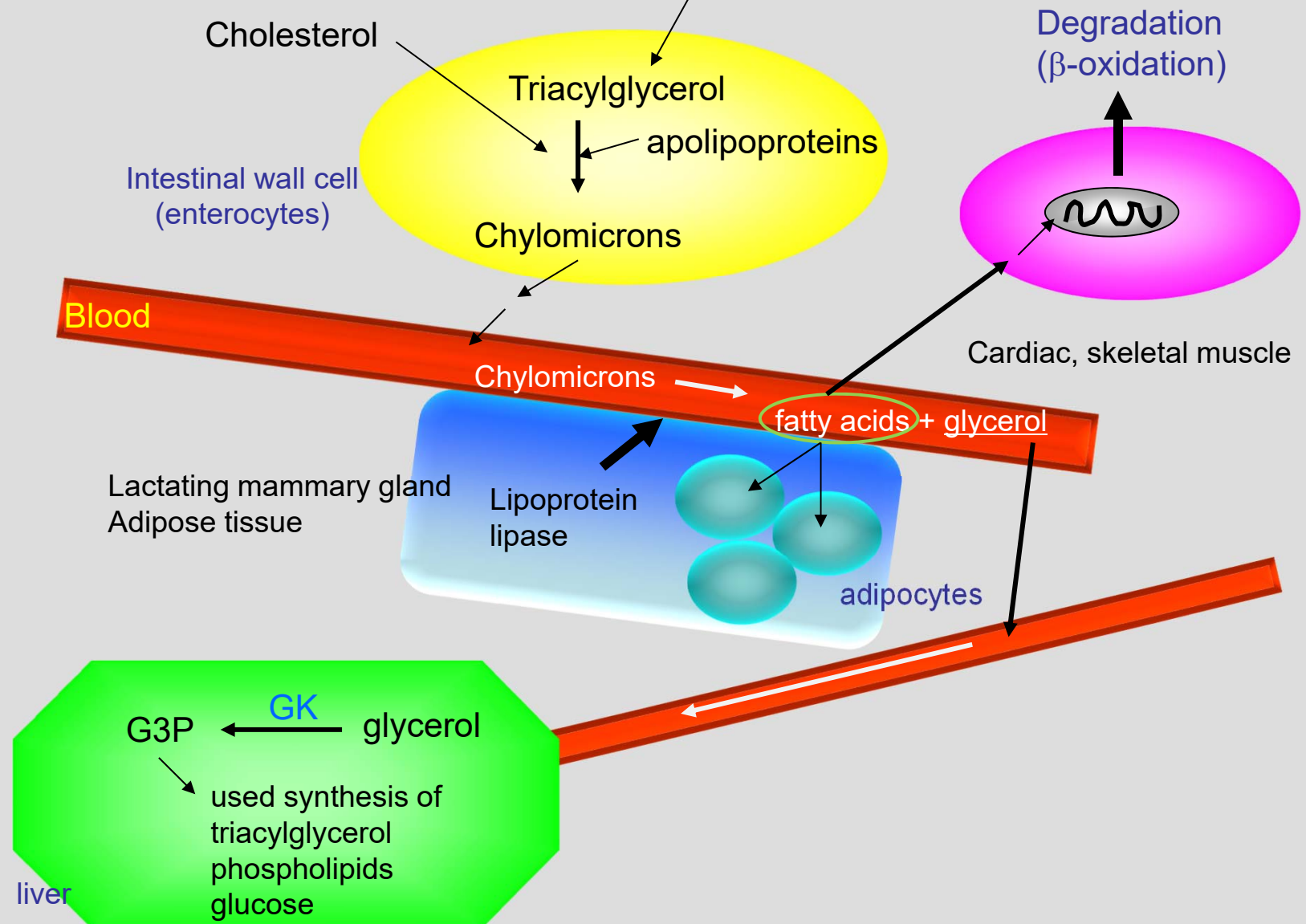
Palmitic acid : 106 ATP/ 16 carbon = 6.6 ATP/carbon

Glucose : 31 ATP/ 6 carbon = 5.2 ATP/carbon

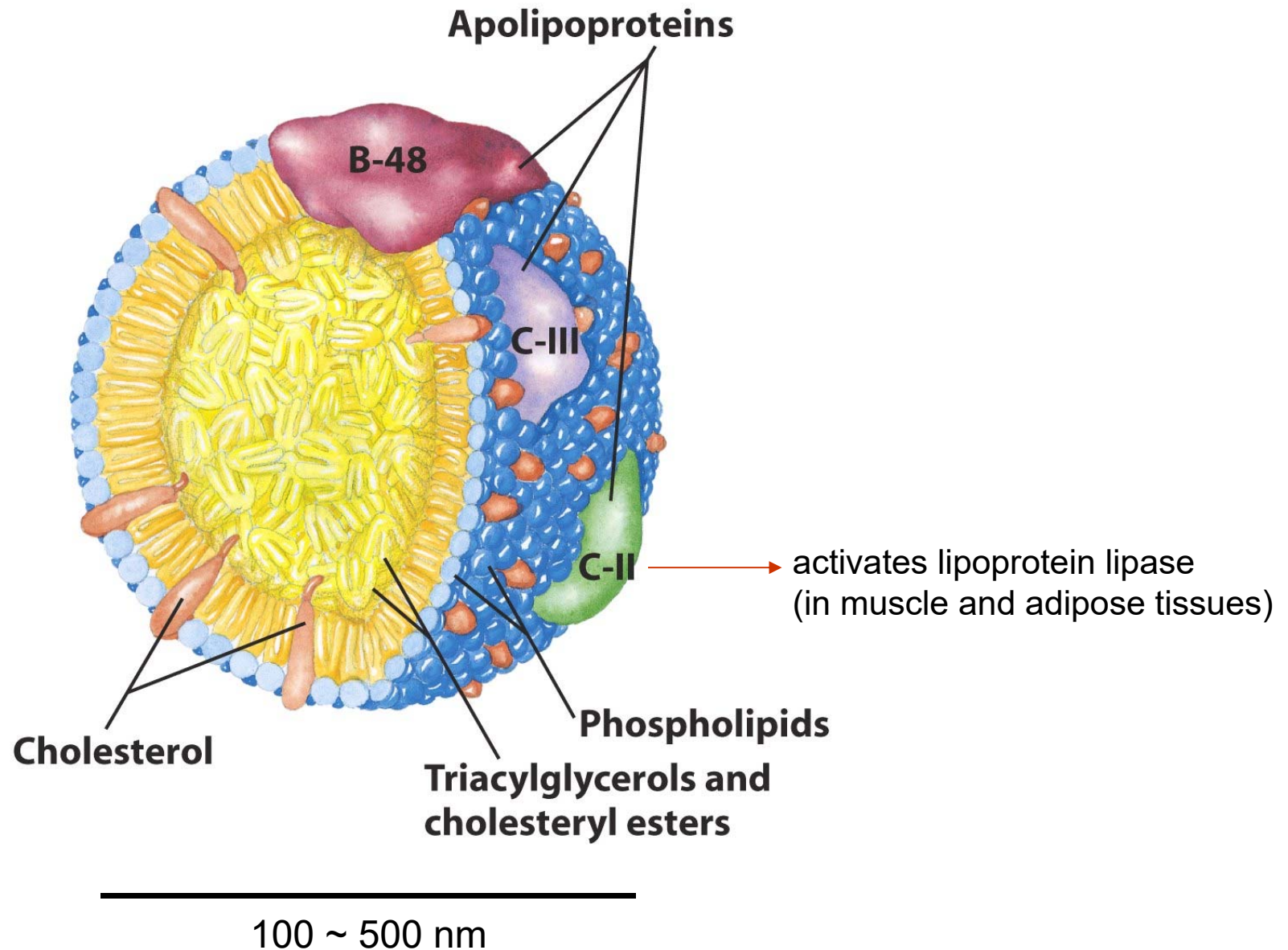
Processing of dietary lipids in vertebrates



Triacylglycerol $\xrightarrow{\text{lipase}}$ fatty acid + glycerol



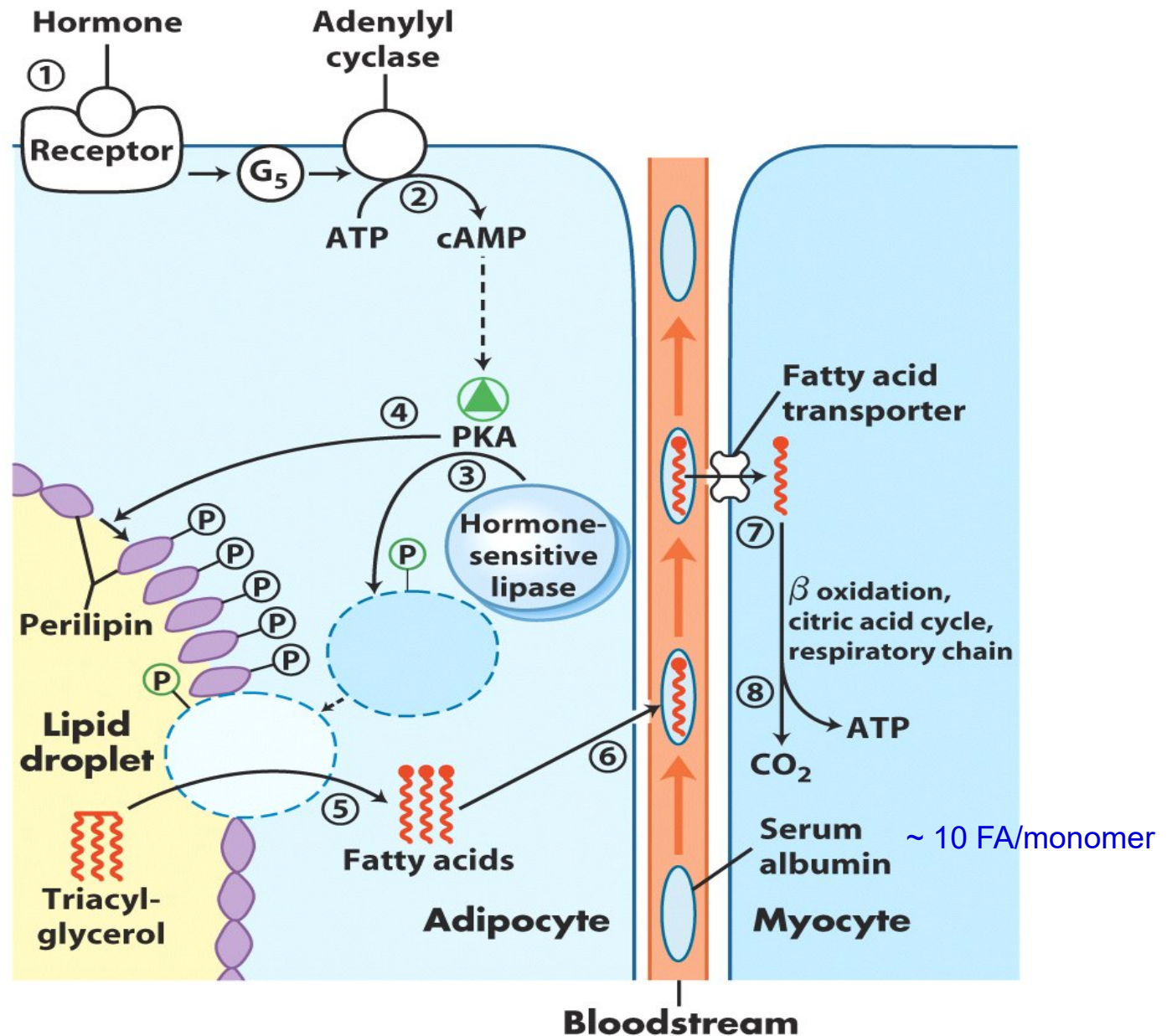
Molecular structure of a chylomicron



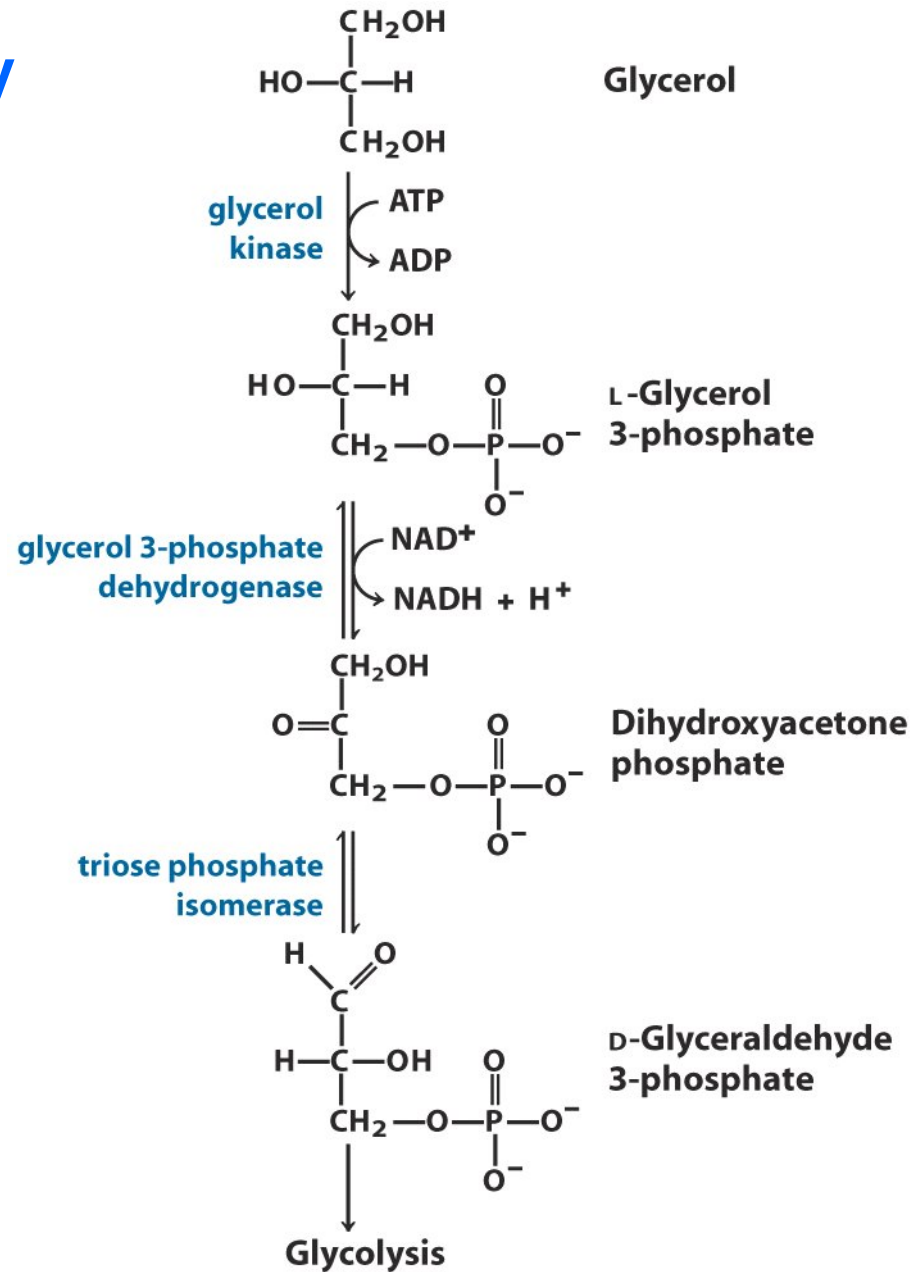
Mobilization of TAGs stored in adipose tissue

Serum glucose level (H):
[insulin]
TAG synthesis (lipogenesis)

Serum glucose level (L):
[glucagon, epinephrine]
TAG breakdown (lipolysis)

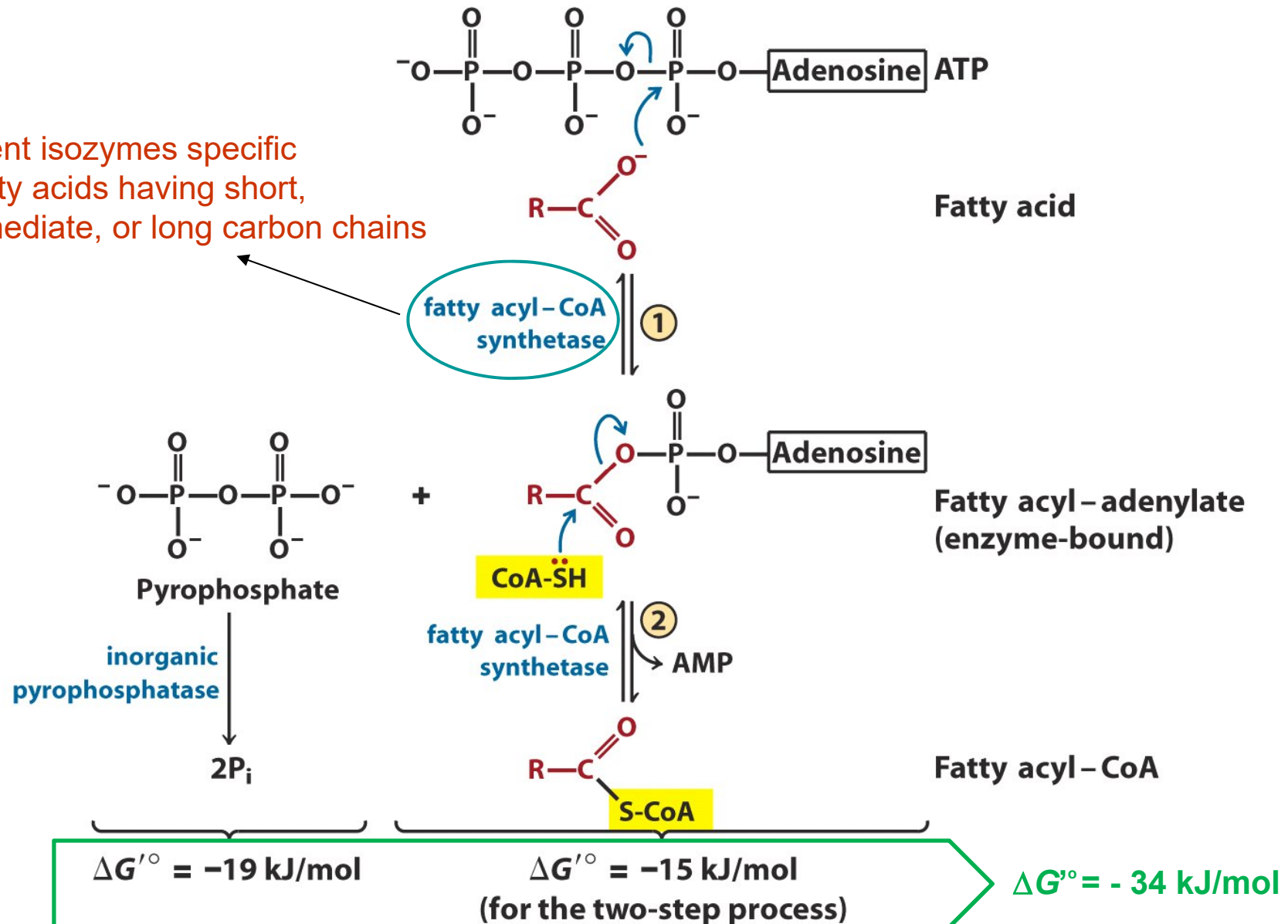


Entry of **glycerol** into the glycolytic pathway

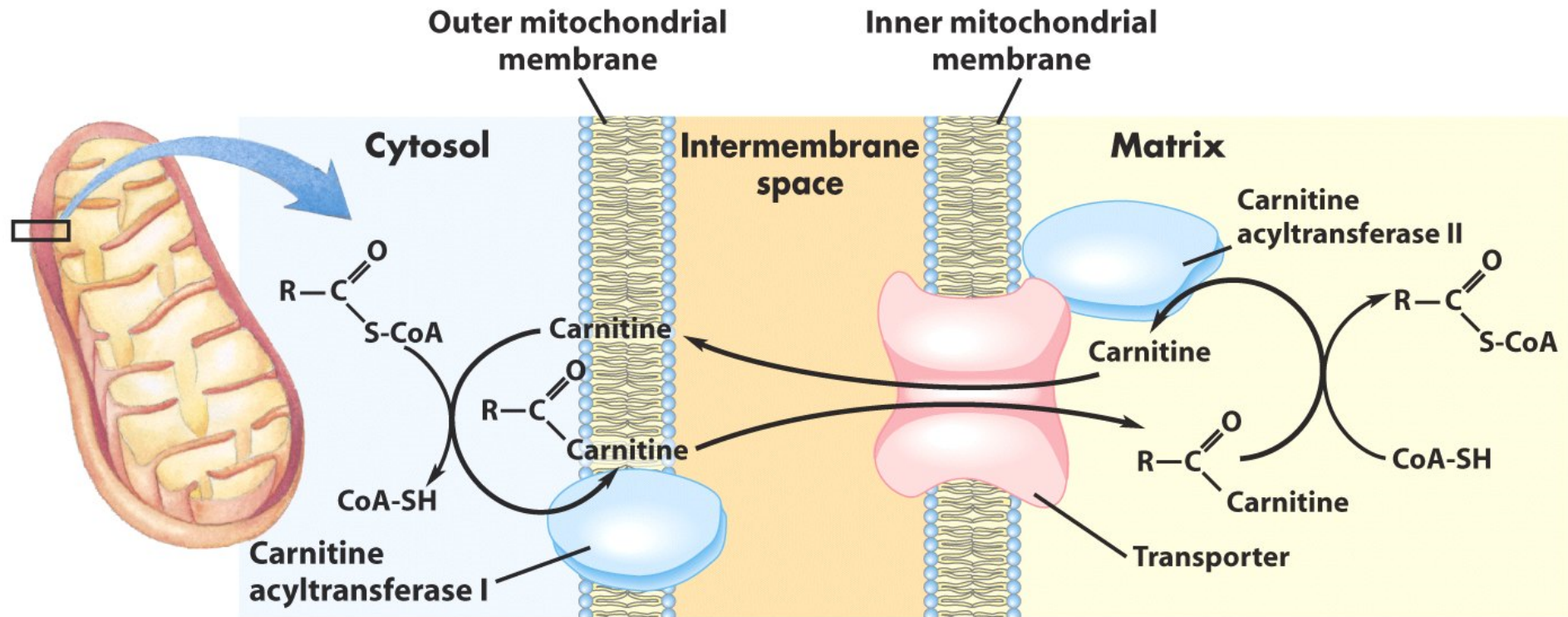


Conversion of fatty acid to a fatty acyl-CoA

different isozymes specific for fatty acids having short, intermediate, or long carbon chains



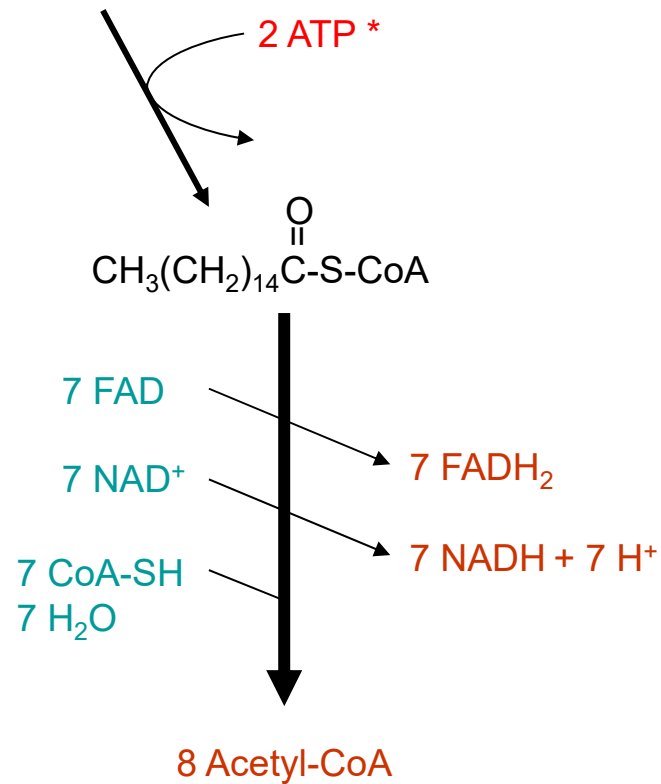
Fatty acids (> 14 carbon) entry into mitochondria via the acyl-carnitine/carnitine transporter



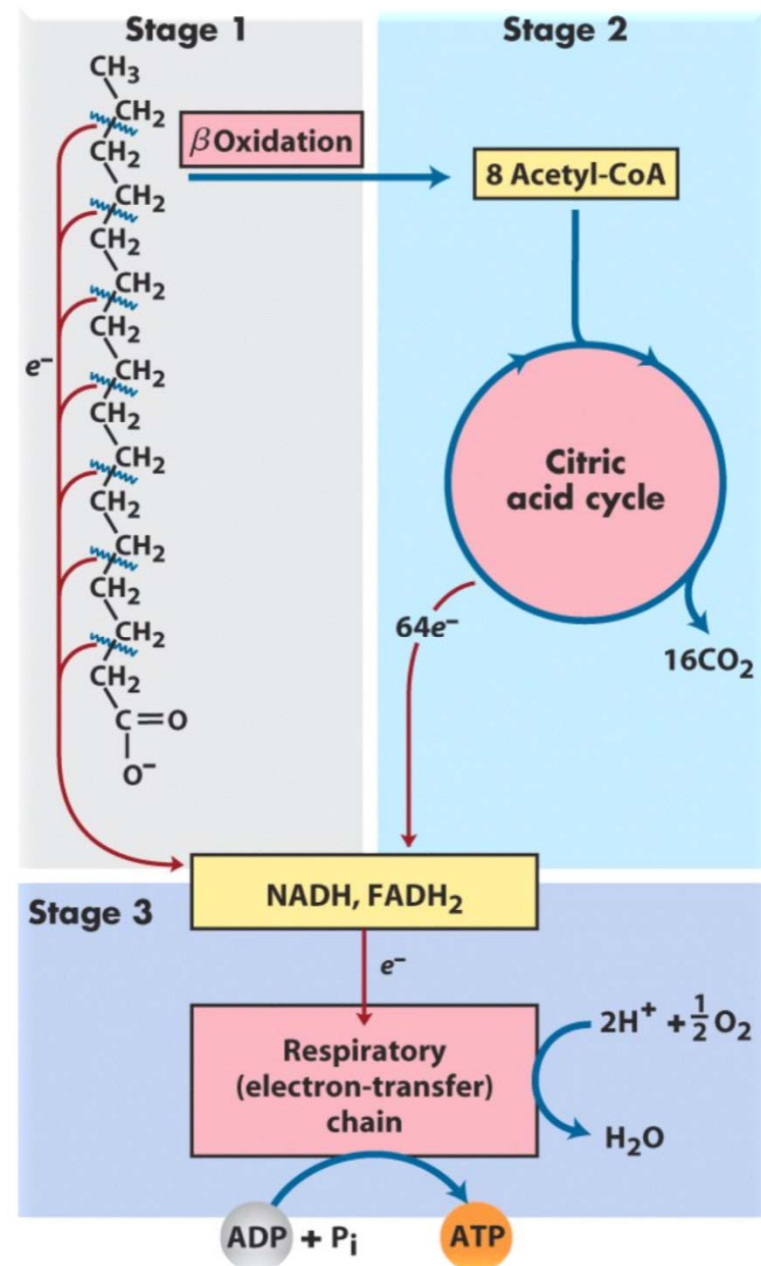
Fatty Acids
(12 or fewer carbons)



Stages of fatty acid oxidation (eg. Palmitic acid)



7 FADH_2	x	1.5 ATP/ FADH_2	=	10.5 ATP
7 NADH	x	2.5 ATP/ NADH	=	17.5 ATP
8 acetyl-CoA	x	10 ATP/acetyl-CoA	=	80 ATP
				108 ATP
				- 2 ATP *
				106 ATP

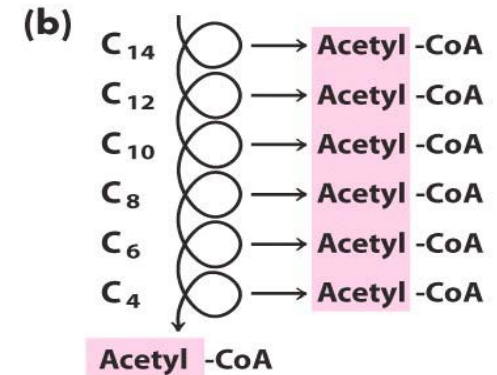
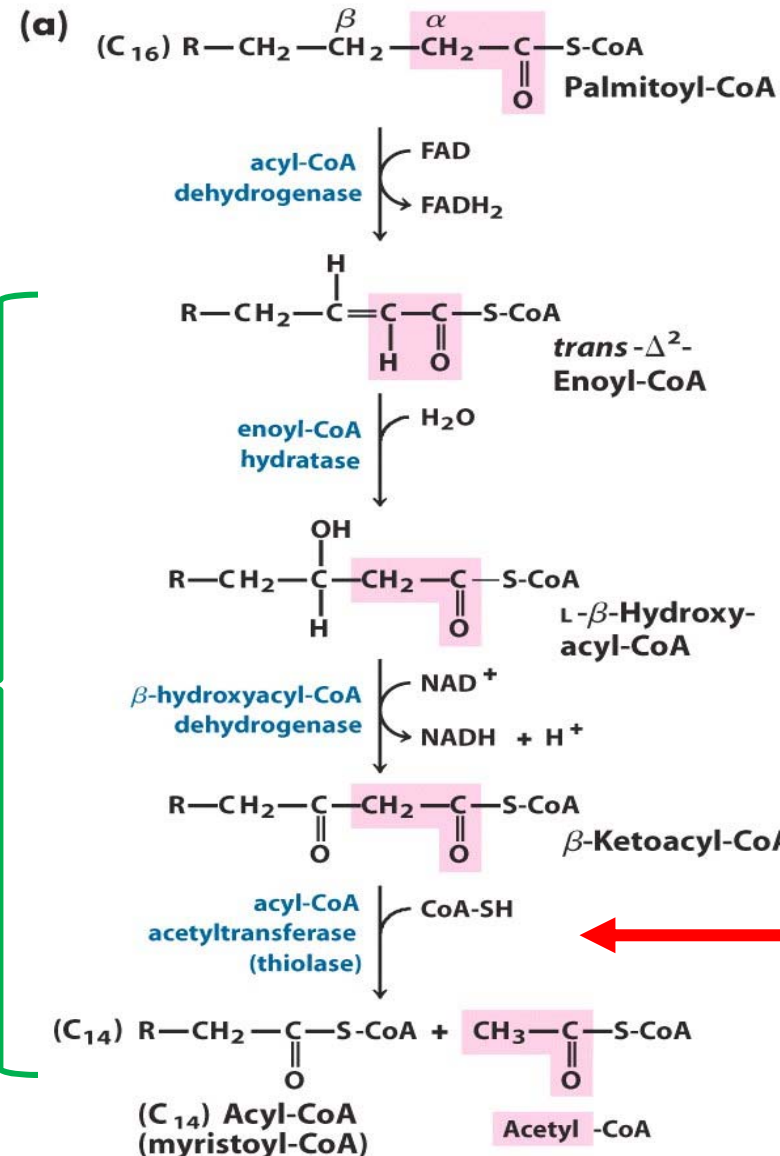


The β -oxidation pathway (of saturated fatty acids)

Two sets of enzymes

TFP complex ($\alpha_4\beta_4$)
(> 12 carbons)

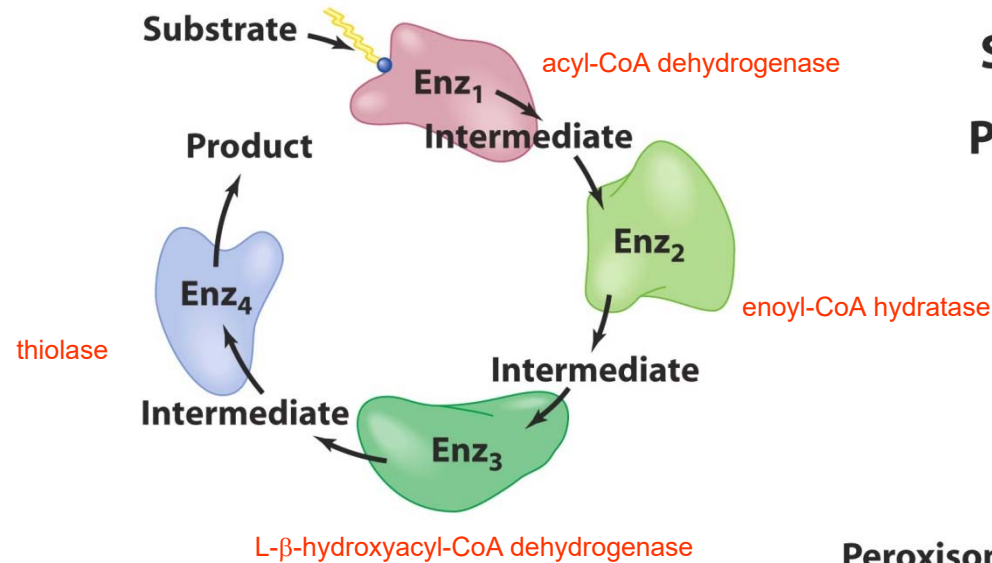
Soluble enzymes
(< 12 carbons)



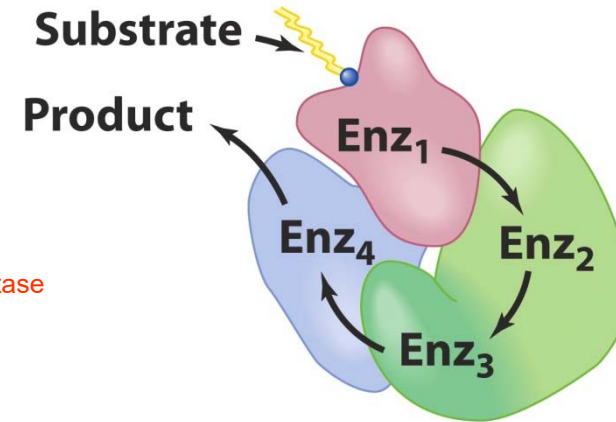
thiolysis

The enzymes of β -oxidation

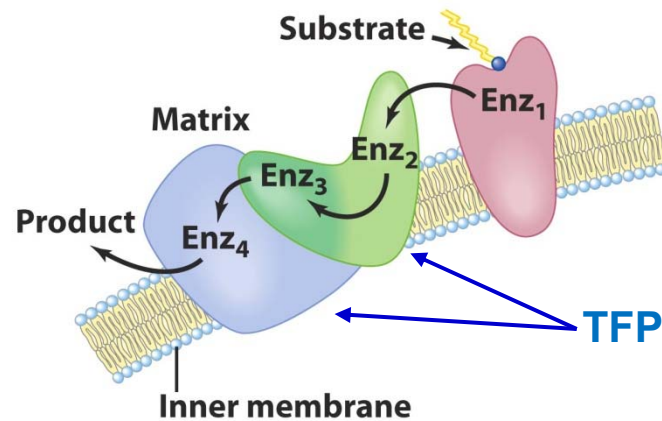
Gram-positive bacteria and mitochondrial short-chain-specific system



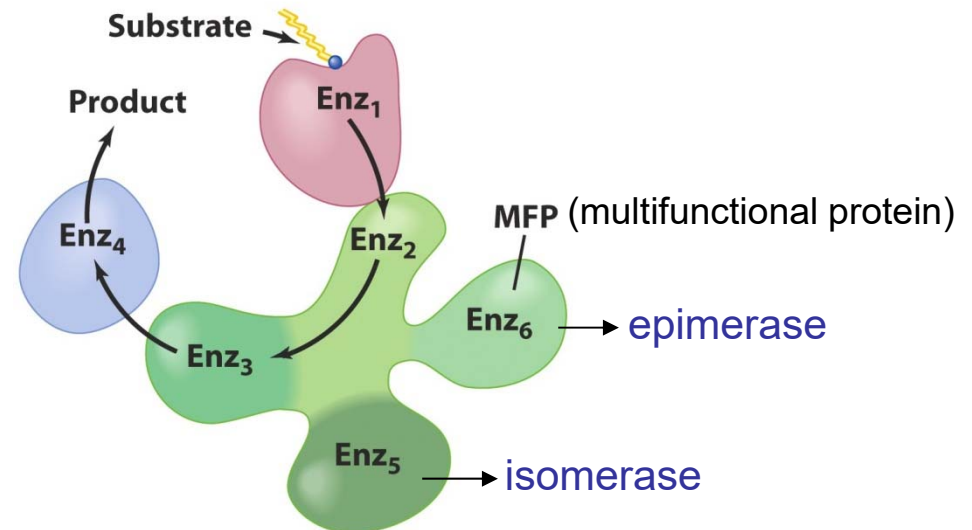
Gram-negative bacteria



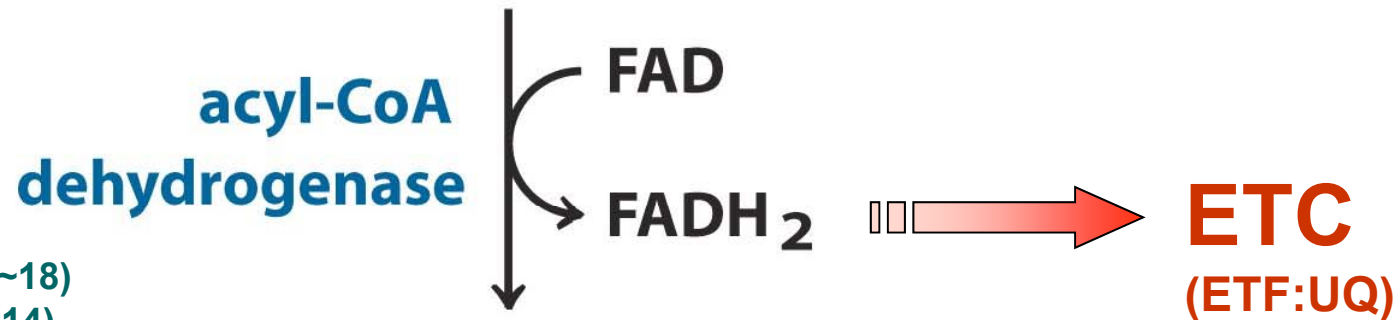
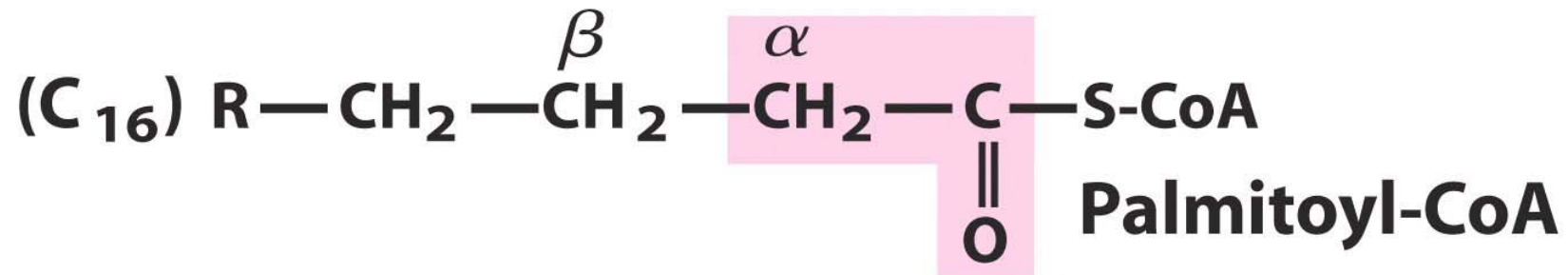
Mitochondrial very-long-chain-specific system



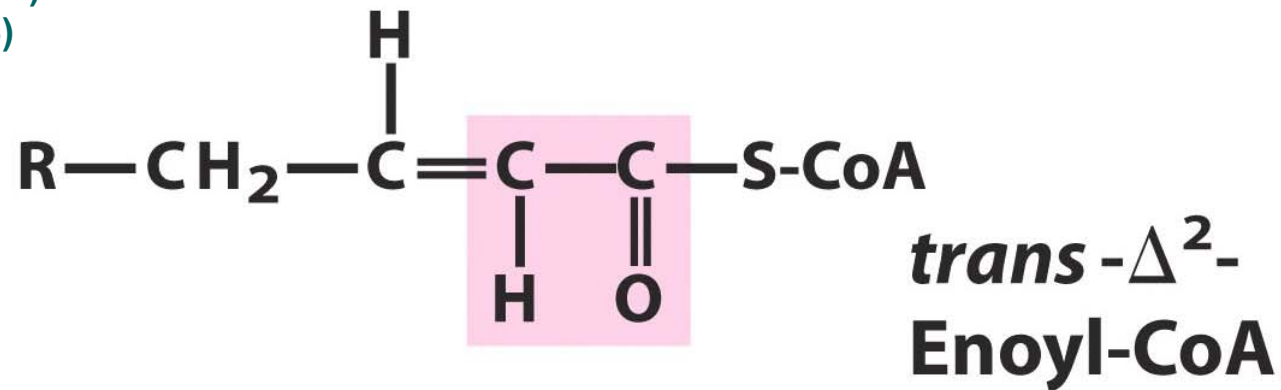
Peroxisomal and glyoxysomal system of plants



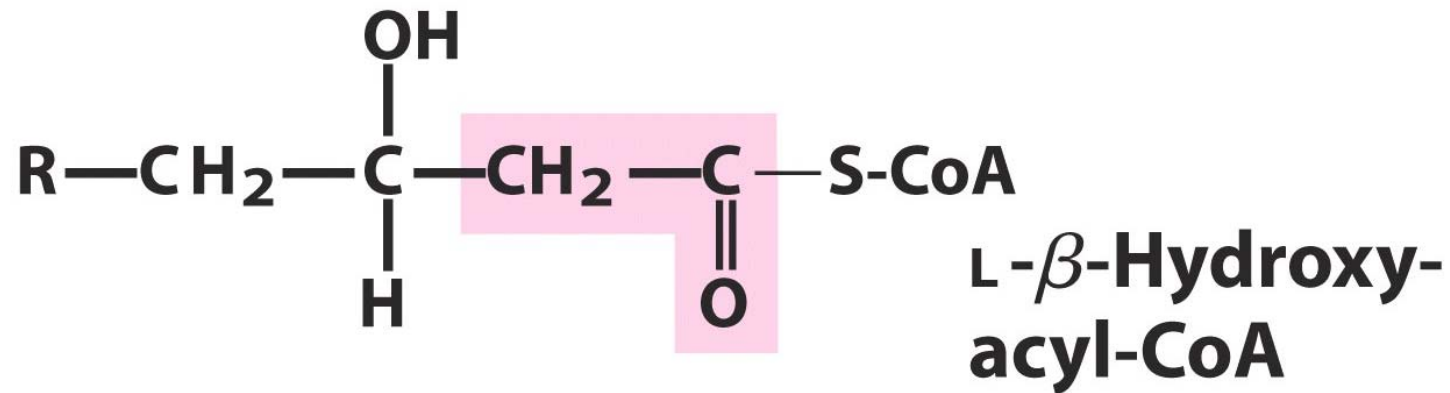
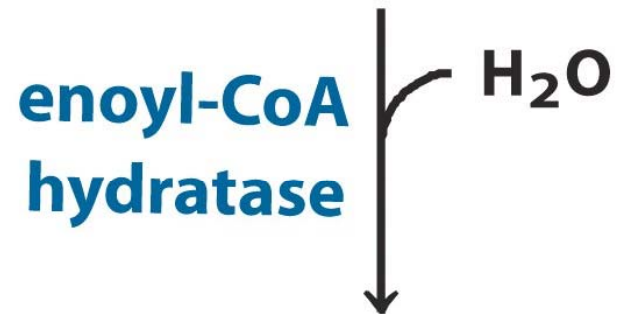
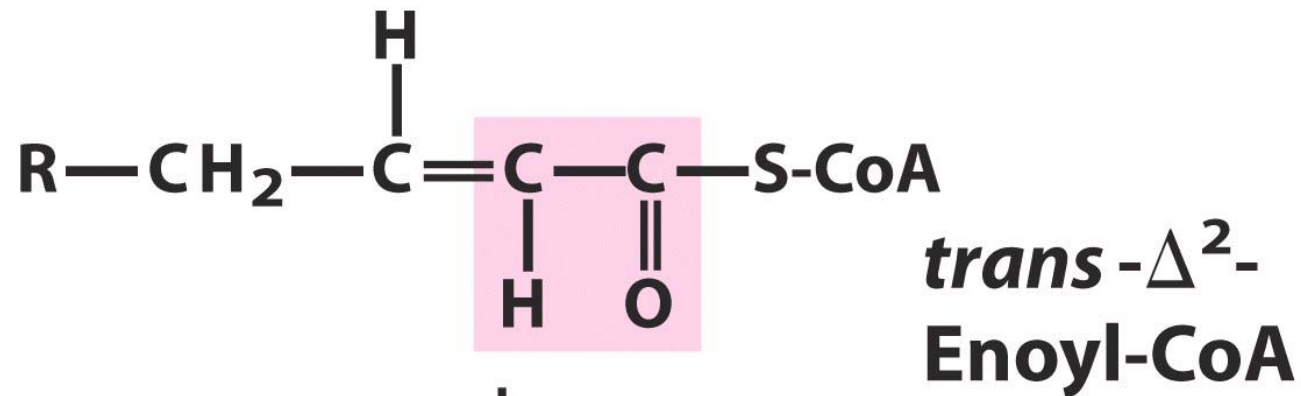
Step 1



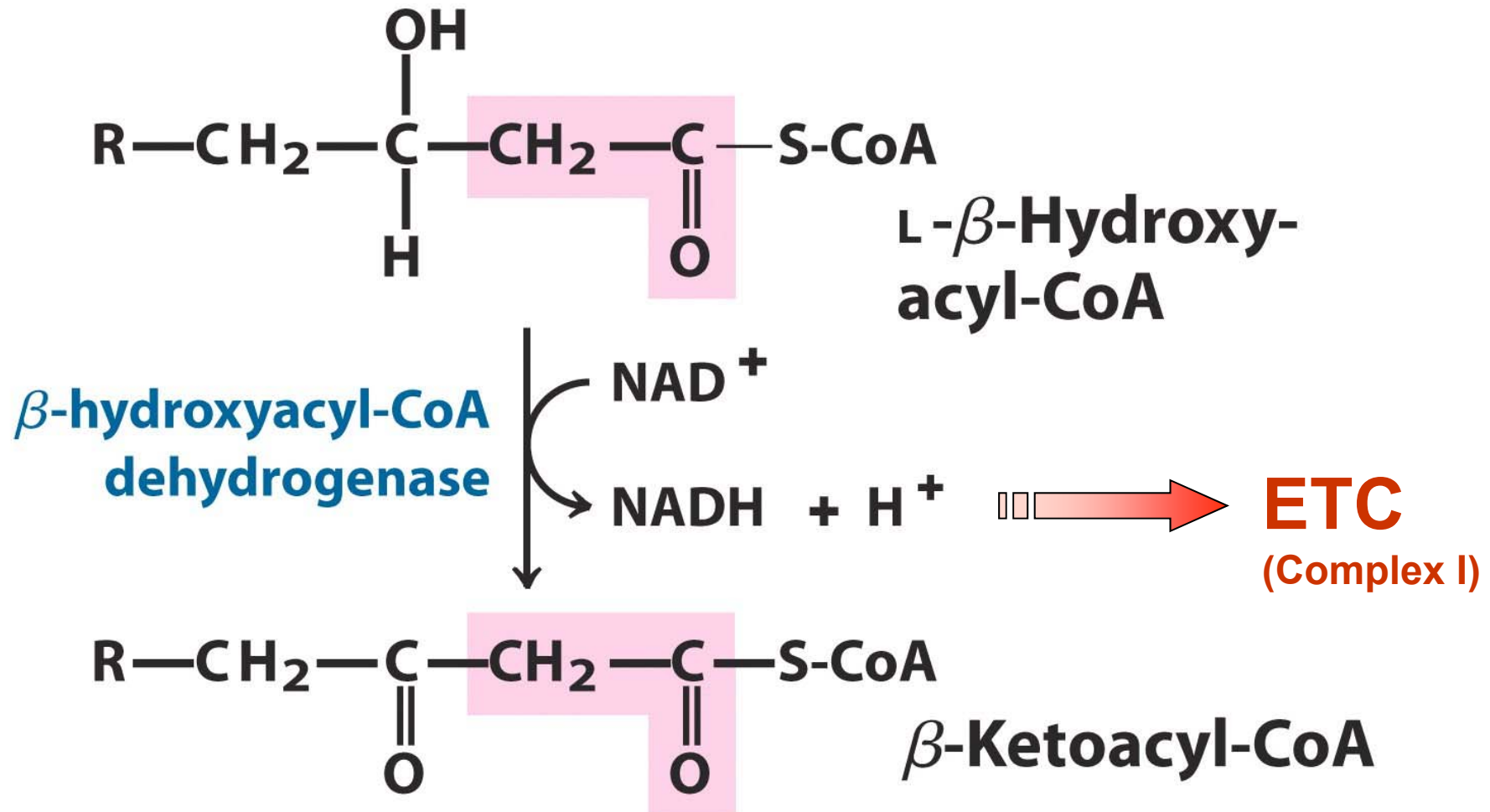
VLCAD (12 ~ 18)
MCAD (4 ~ 14)
SCAD (4 ~ 8)



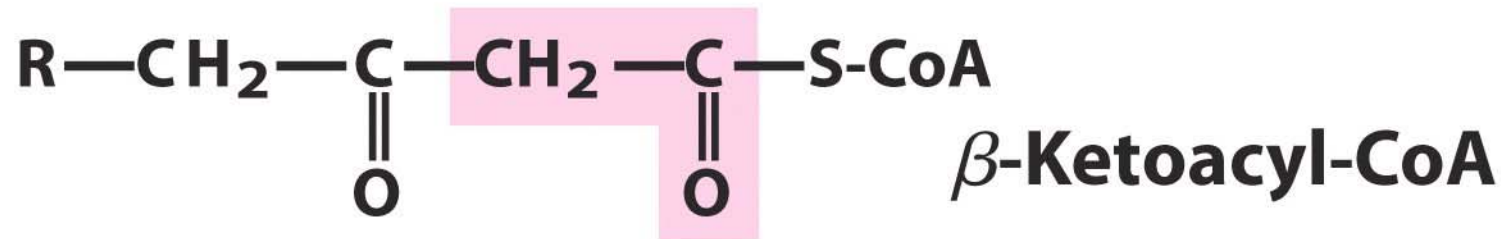
Step 2



Step 3

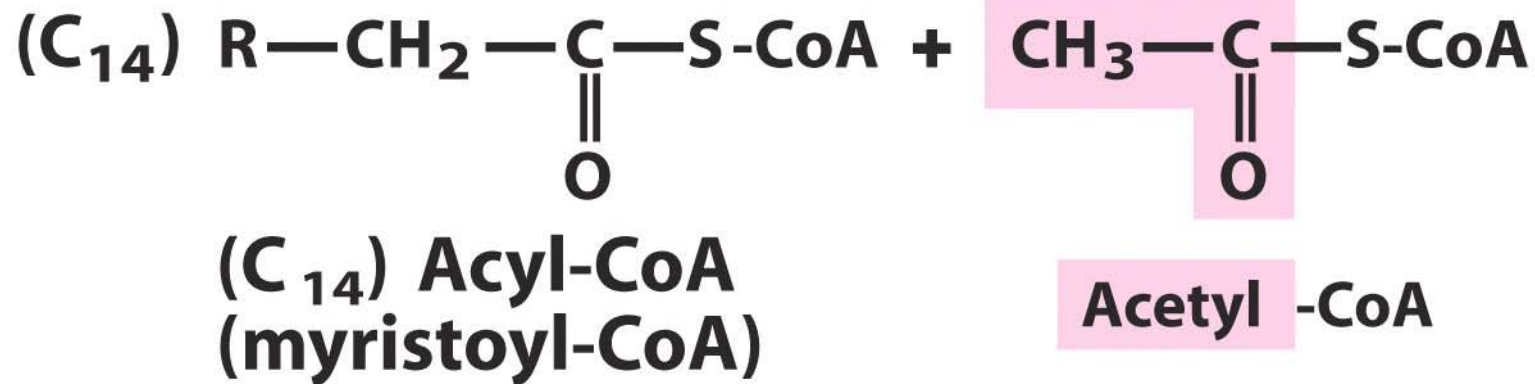


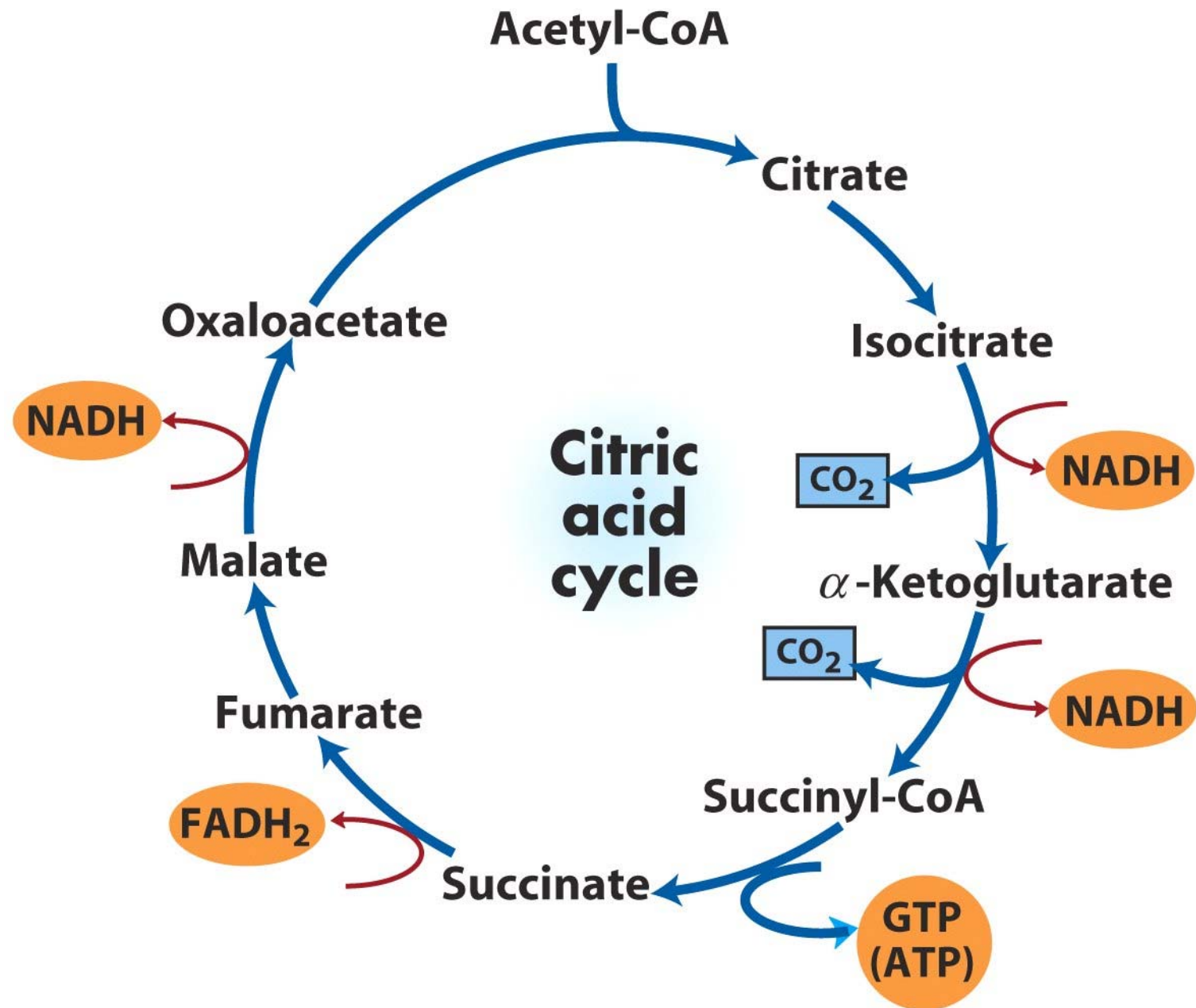
Step 4



acyl-CoA
acetyltransferase
(thiolase)

CoA-SH

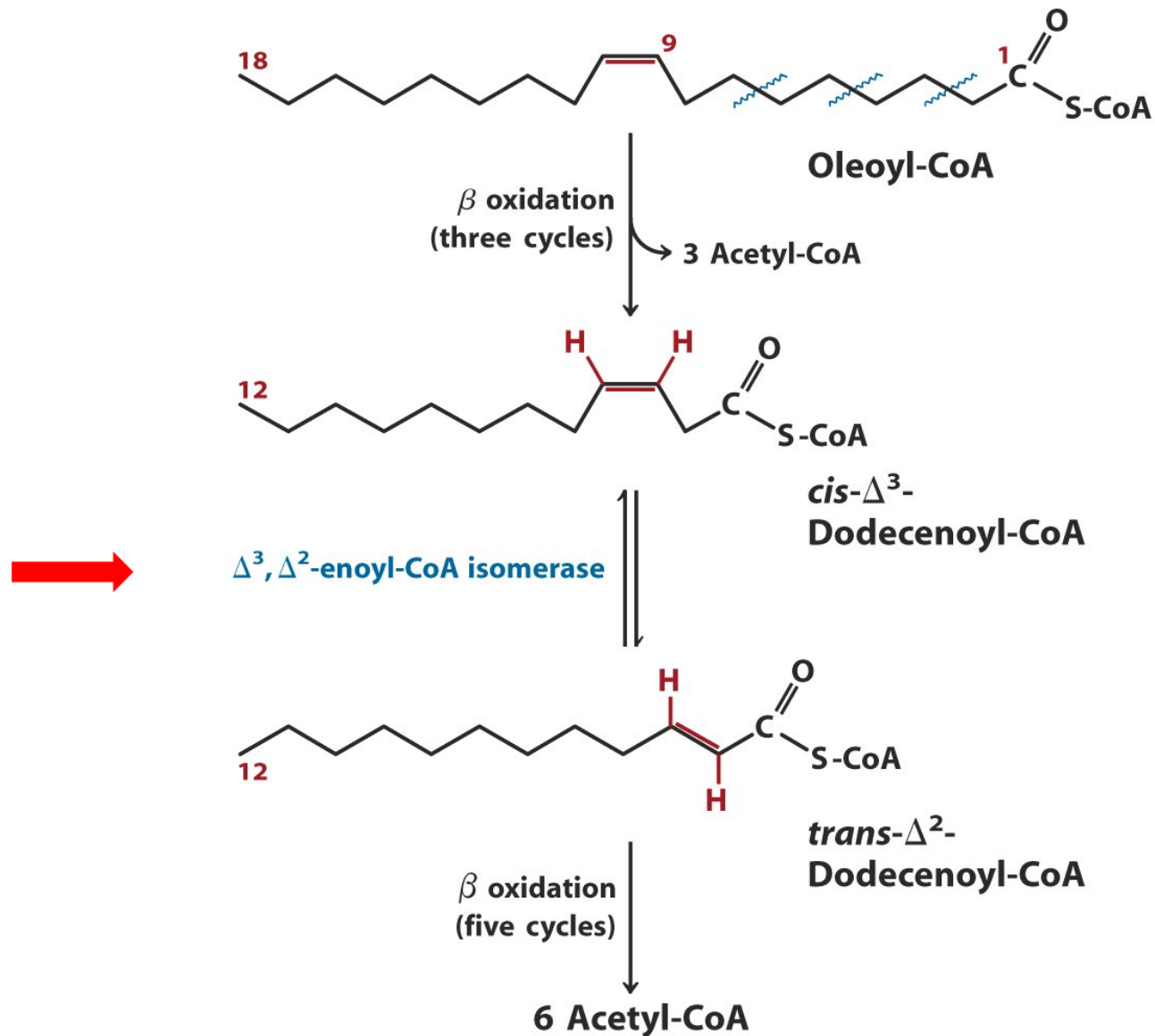




Fat bears carry out β -oxidation in their sleep

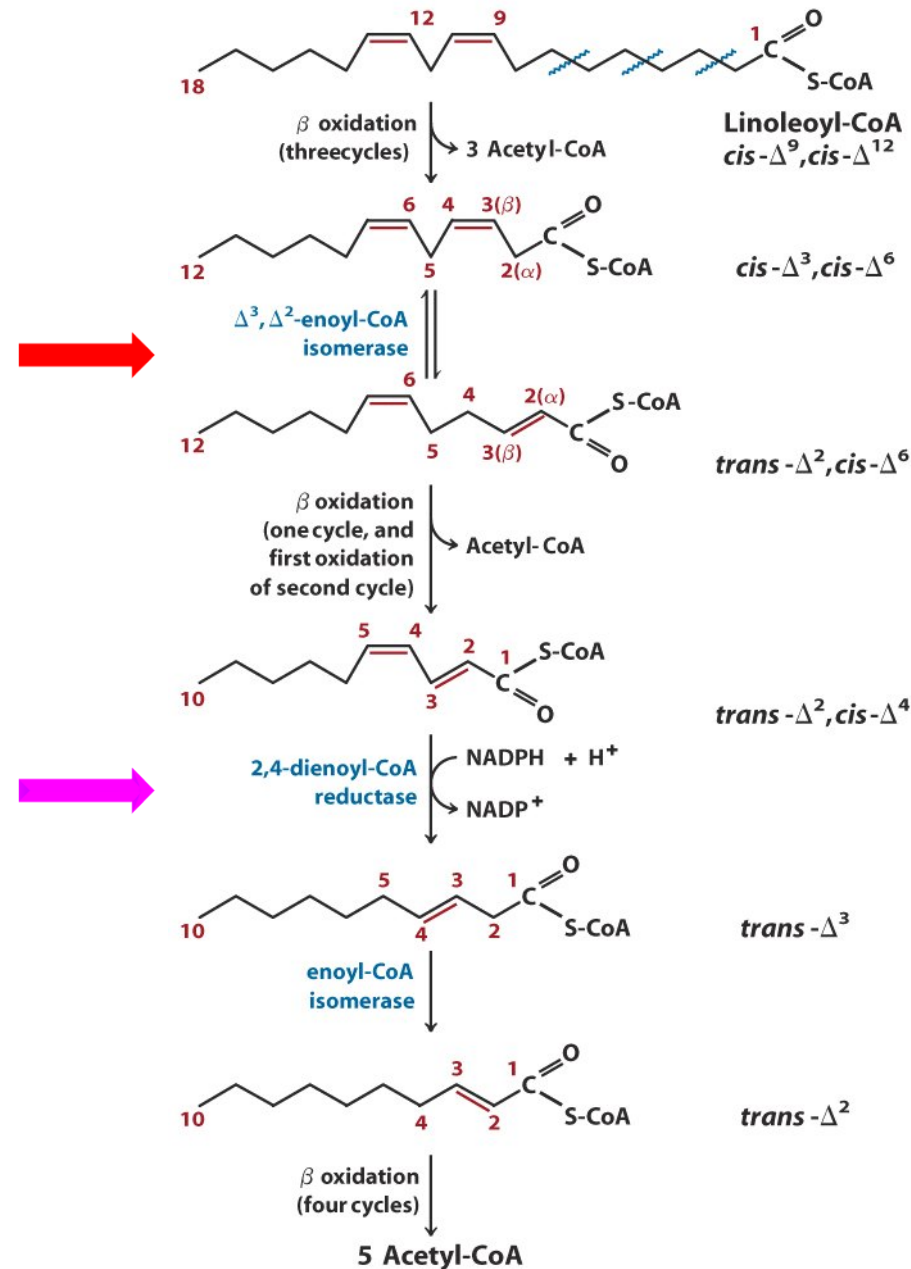


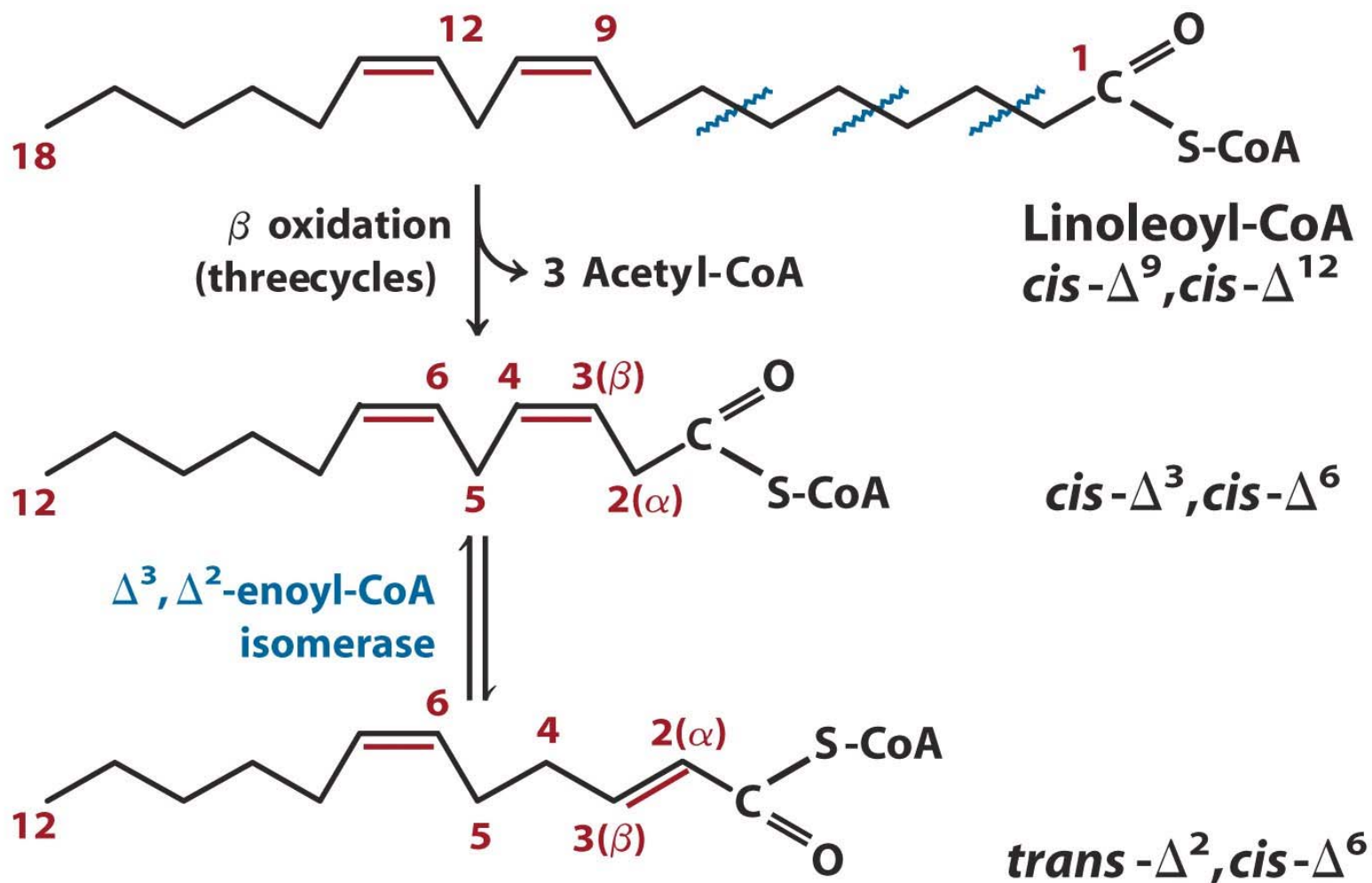
Oxidation of monounsaturated fatty acid

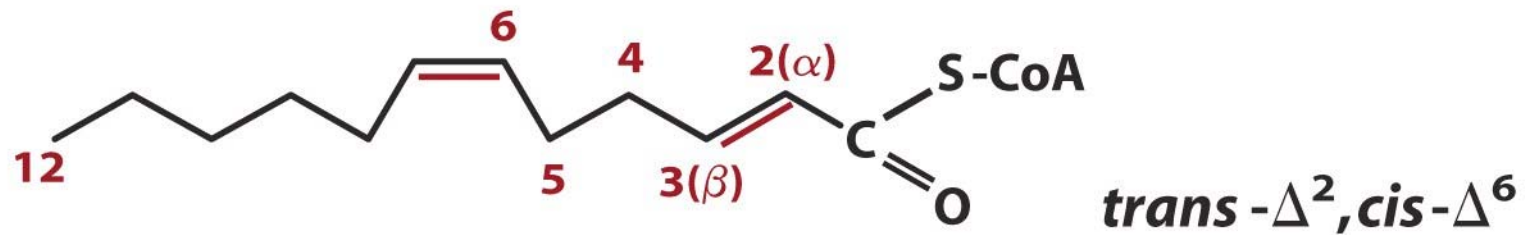


Oxidation of polyunsaturated fatty acid

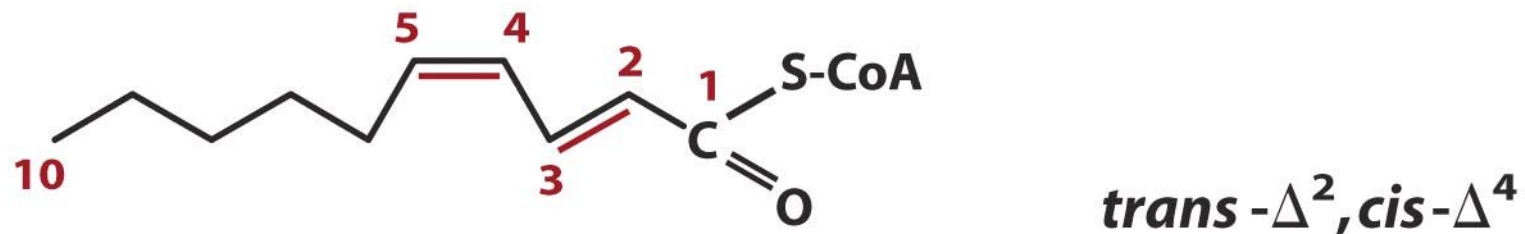
(Linoleic acid, 18:2 $\Delta^{9,12}$)





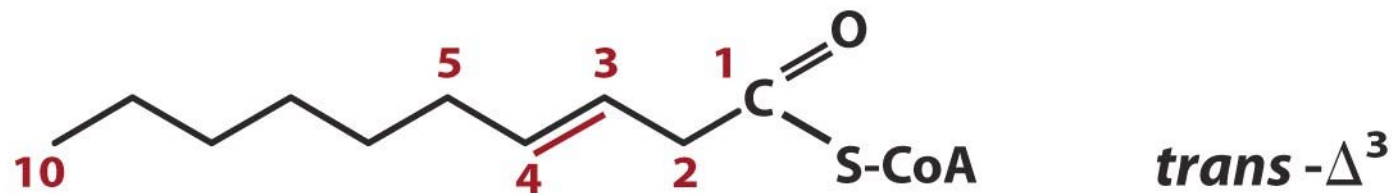


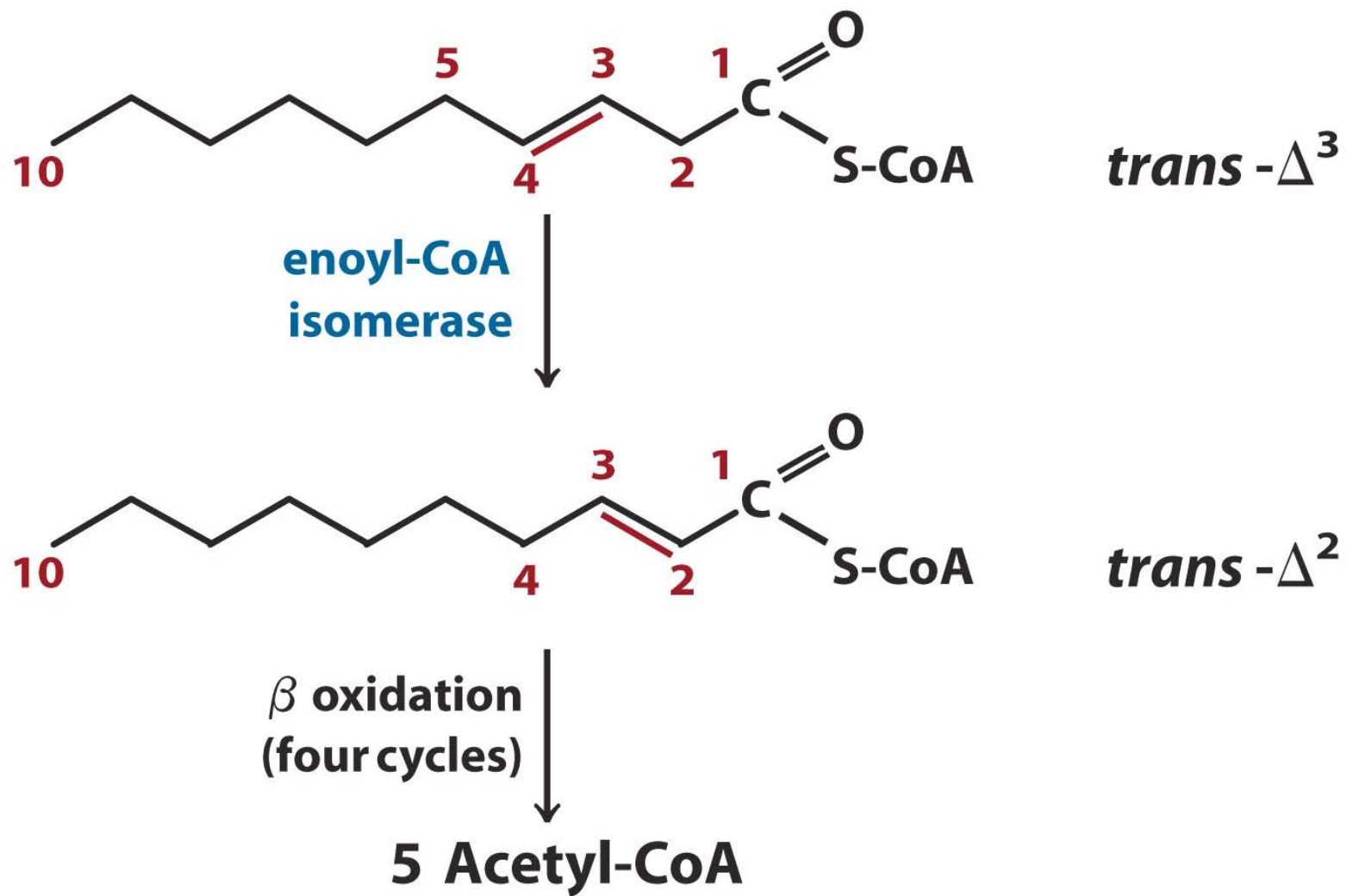
β oxidation
 (one cycle, and
 first oxidation
 of second cycle)



2,4-dienoyl-CoA
 reductase

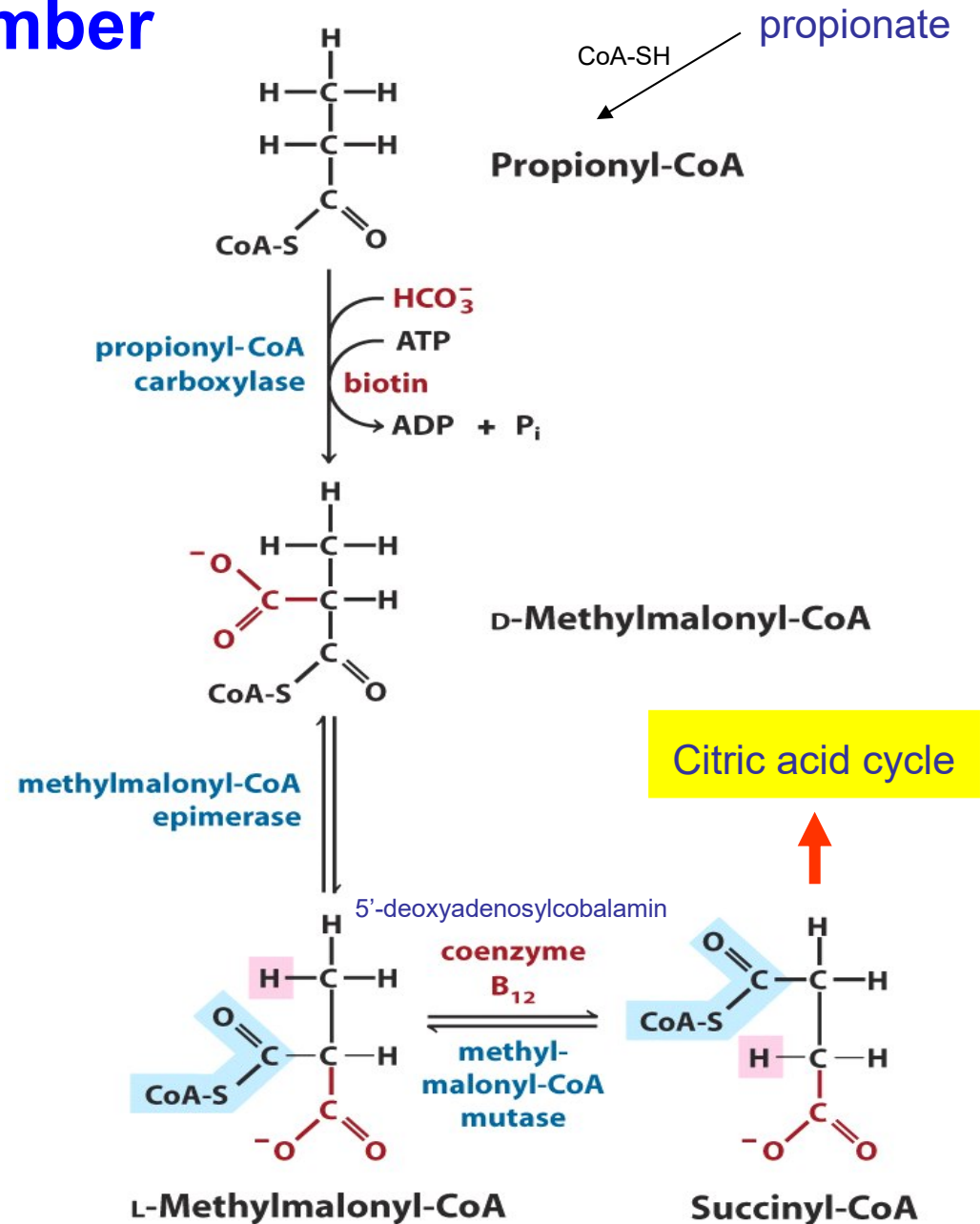
NADPH + H⁺
 NADP⁺





Oxidation of odd-number fatty acid

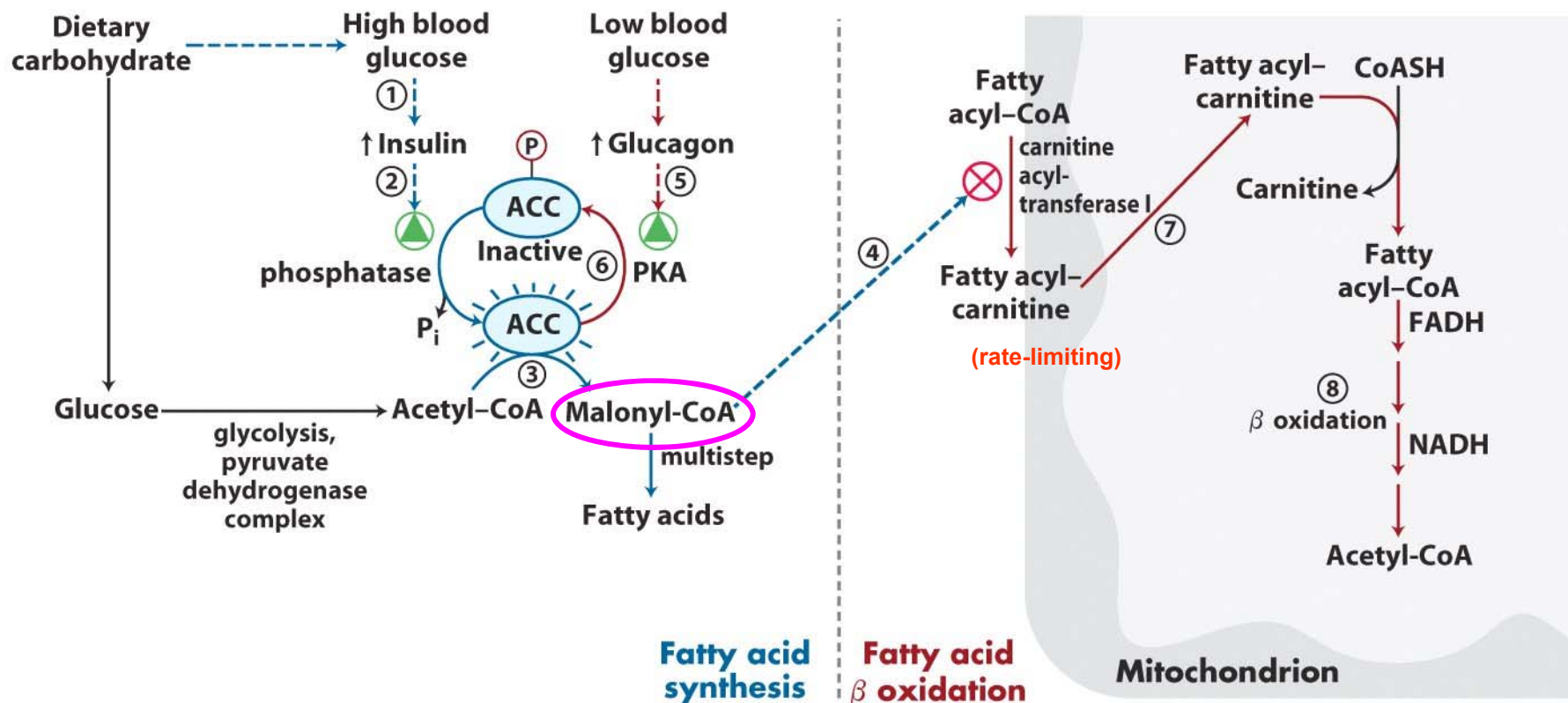
(in the lipids of many plants and some marine organisms)



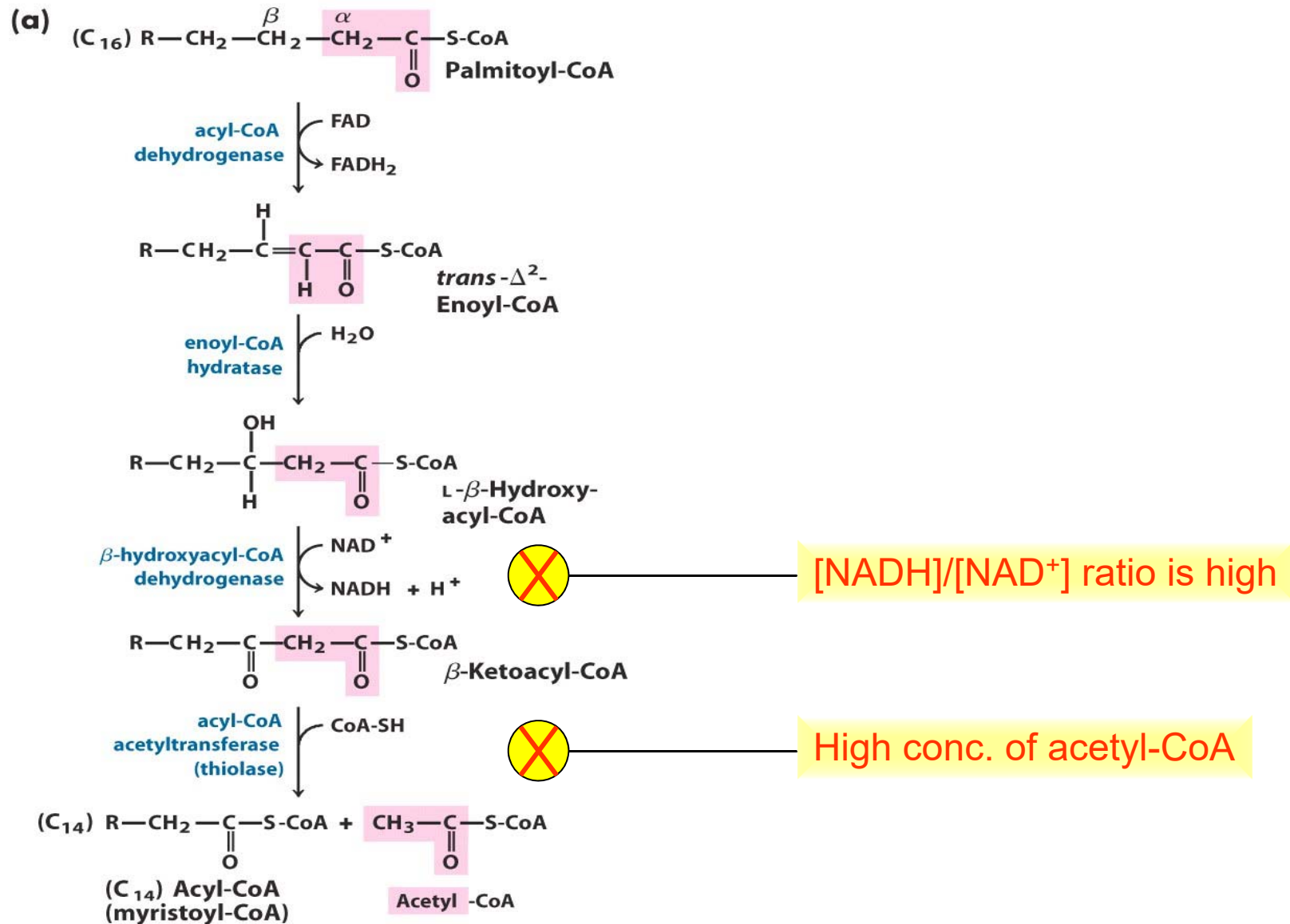
* Pernicious anemia (intrinsic factor ↓)

Coordinated regulation of fatty acid synthesis and breakdown

TAG or phospholipids ← **Fatty acyl-CoA** → **β-oxidation**

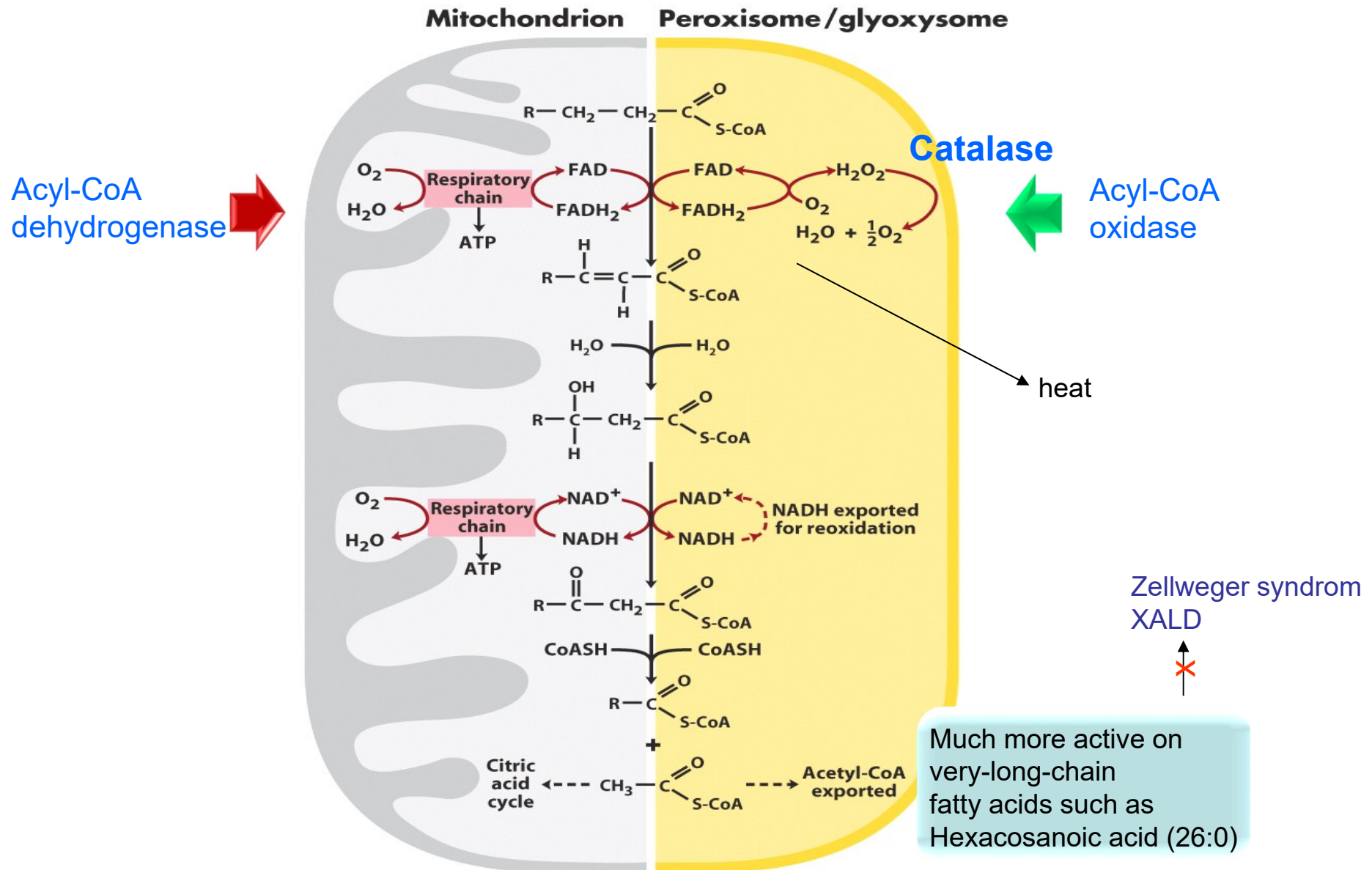


Two inhibition steps for β -oxidation



Peroxisome also carry out β -oxidation

(major site of β -oxidation in plant)



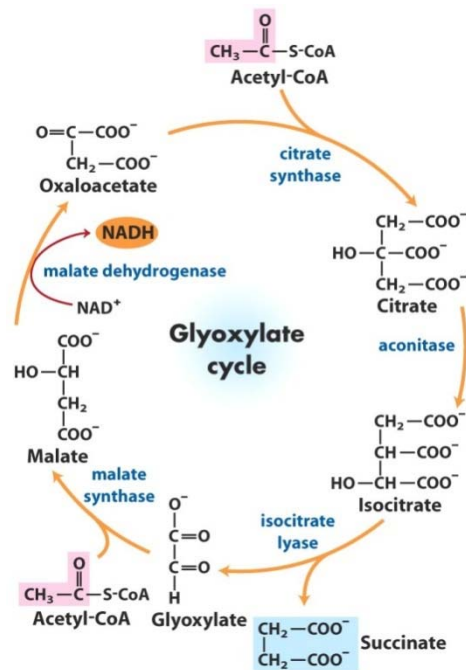
Triacylglycerols as glucose source in seeds

Fatty acid oxidation in plants:

not in mitochondria

but in the peroxisomes

(and the glyoxysomes of
germinating seeds)



Seed triacylglycerols

lipases

Fatty acids

β oxidation

Acetyl-CoA

glyoxylate cycle

Oxaloacetate

gluconeogenesis

Glucose

Sucrose,
polysaccharides

Metabolic
intermediates

Amino acids

Nucleotides

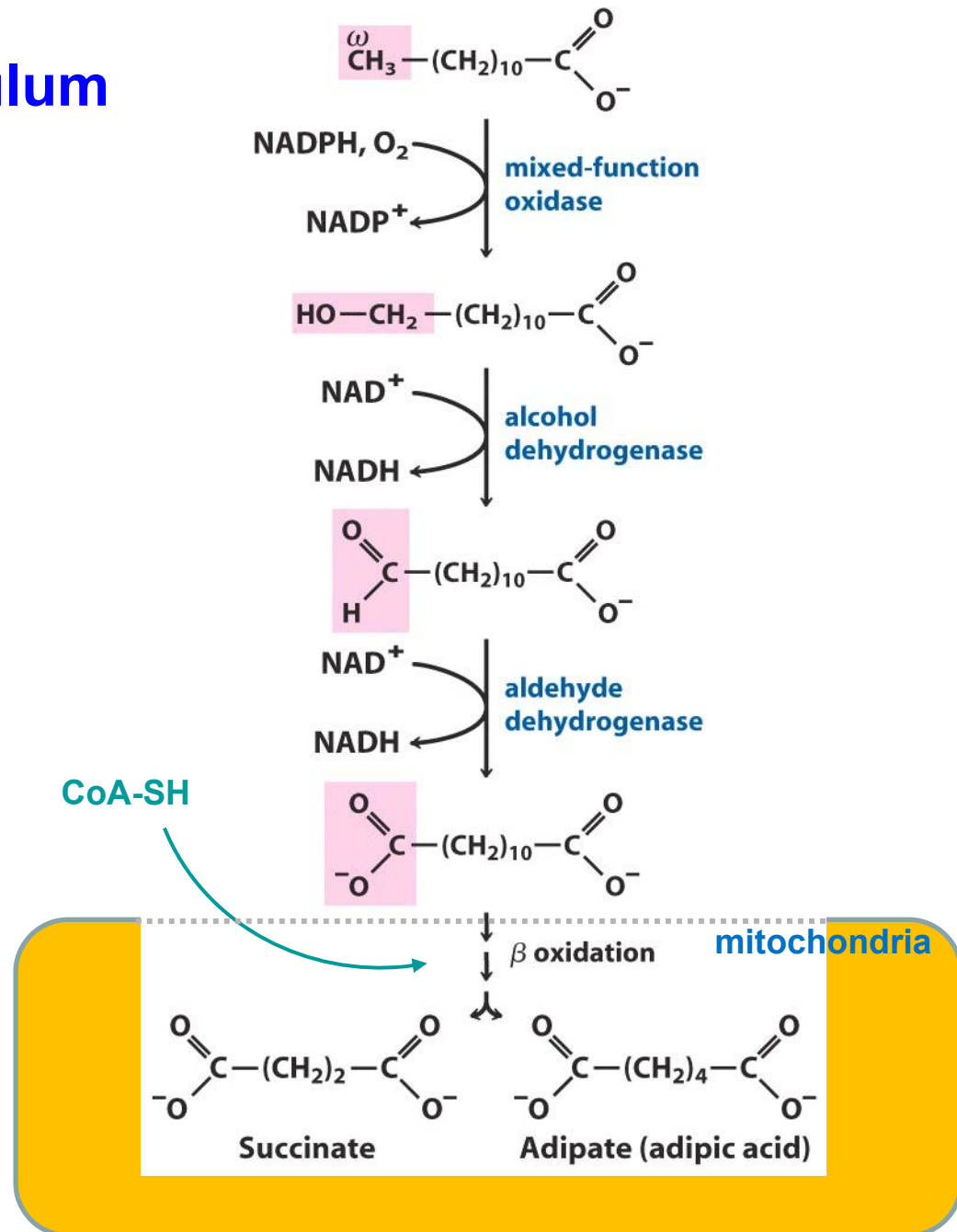
Energy

ω -oxidation in the endoplasmic reticulum (liver and kidney in animal)

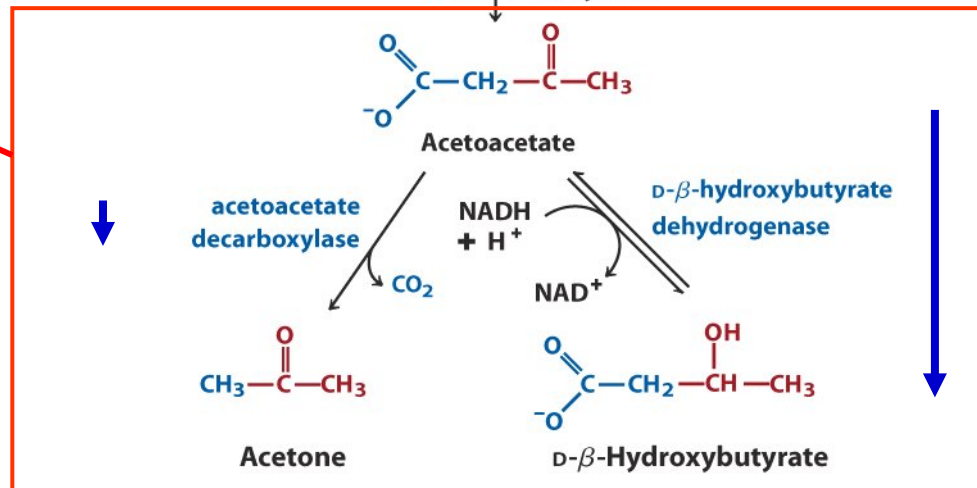
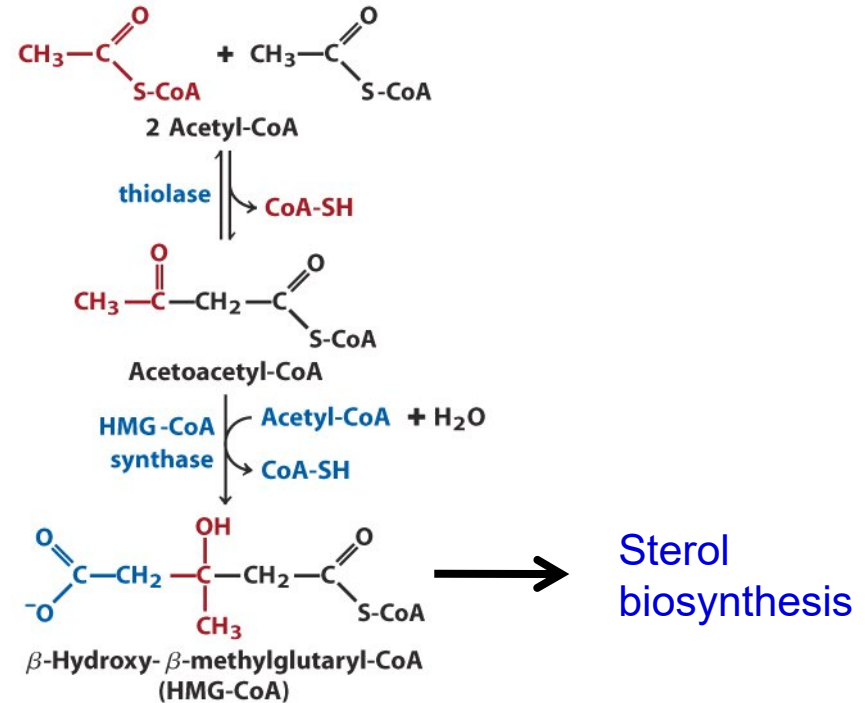
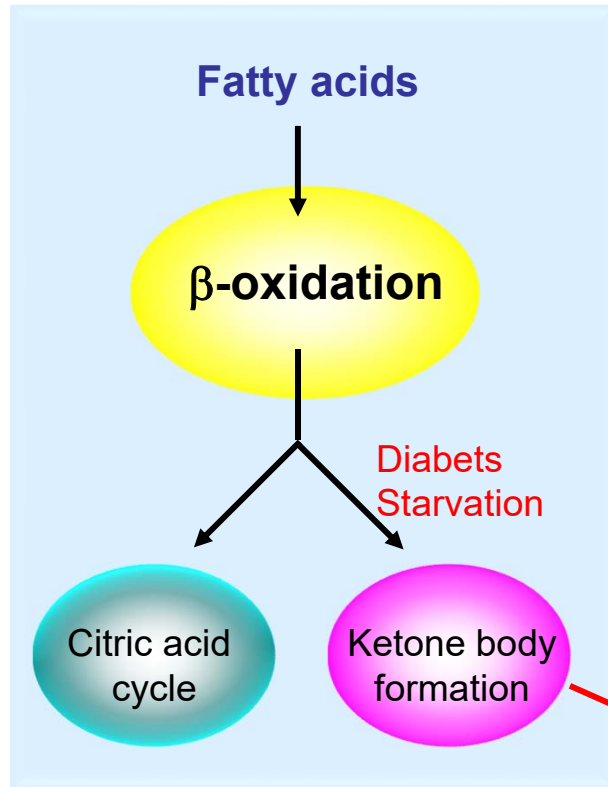
(10:0 or 12:0)

α -oxidation:

- branched-chain fatty acids
- in peroxisome of animal cells

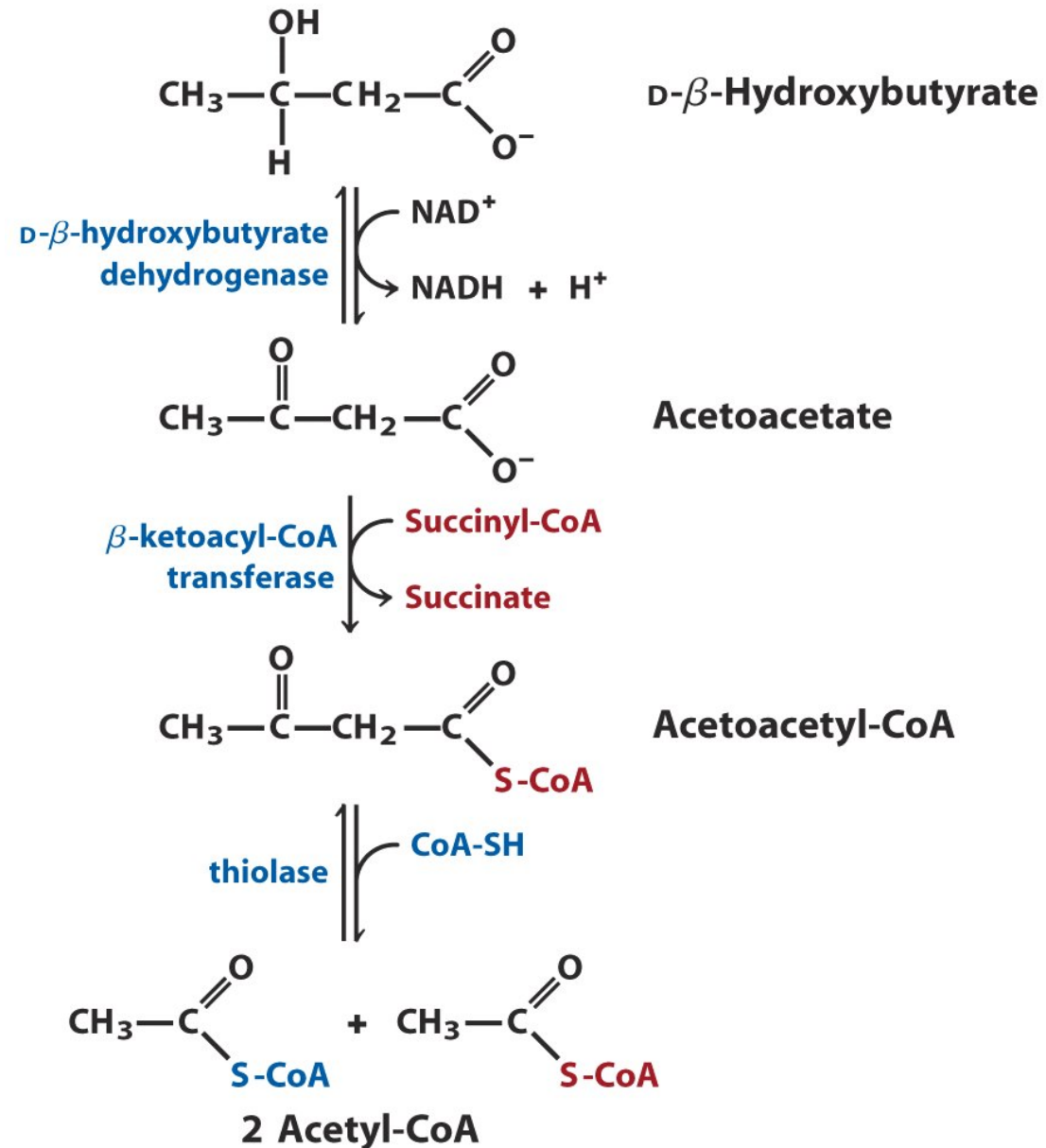


Formation of ketone bodies from acetyl-CoA



β -Hydroxybutyrate as a fuel

(in extrahepatic tissues)



Ketone body formation and export from the liver

Acidosis

Ketosis

90 mg/100 ml in blood
(<3 mg/100 ml)
5000 mg/24 h urinary excretion
(<125 mg/24 h)

